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## INCREASING THE EFFICIENCY OF CRYOGENIC AUTOMOBILE POWER SYSTEMS USING THERMOELECTRIC GENERATORS

*The application of the recently discovered low-temperature thermoelectric material CsBi<sub>4</sub>Te<sub>6</sub> in thermoelectric generators (TEG) for automotive cryogenic power systems is proposed. The maximum energy conversion efficiency of a considered TEG assembly within a cryogenic storage tank is estimated to be about 15%. To determine specific power, heat flow through the TEG was calculated using a one-dimensional thermal model. It has been obtained that low-temperature TEGs are applicable for additional power generation in cryogenic power systems, and these generators can sufficiently increase total energy efficiency.*

*Предложено применение недавно открытых низкотемпературных термоэлектрических материалов CsBi<sub>4</sub>Te<sub>6</sub> в термоэлектрических генераторах (ТЭГ) для автомобильных криогенных силовых установок. Максимальная эффективность преобразования энергии в рассмотренной конструкции ТЭГ внутри криогенного резервуара по расчетам составляет около 15%. Для определения удельной мощности тепловой поток через термоэлектрический генератор был рассчитан с использованием одномерной тепловой модели. Было установлено, что ТЭГи применимы для выработки дополнительной энергии в криогенных силовых установках и эти генераторы могут в достаточной степени увеличить общую энергоэффективность.*

*Запропоновано застосування недавно відкритих низкотемпературних термоелектричних матеріалів CsBi<sub>4</sub>Te<sub>6</sub> в термоелектричних генераторах (ТЕГ) для автомобільних криогенних силових установок. Максимальна ефективність перетворення енергії в розглянутій конструкції ТЕГ всередині криогенного резервуара за розрахунками становить близько 15%. Для визначення питомої потужності тепловий потік через термоелектричний генератор був розрахований з використанням одновимірної теплової моделі. Було встановлено, що ТЕГи можуть застосовуватися для вироблення додаткової енергії в криогенних силових установках і ці генератори можуть достатньо мірою збільшити загальну енергоефективність.*

### Introduction

In the last ten years, new non-polluting cryogenic vehicles operating on liquid nitrogen (LN<sub>2</sub>) were developed in the US at the University of North Texas (UNT) and the University of Washington (UW) and in Ukraine at Kharkov Universities (KU) [1–6]. These vehicles are shown in Fig. 1.

A motivation for developing such vehicles is to enable an environmentally friendly means of transportation that does not use batteries or hydrocarbon fuel. The cryogenic propulsion systems in these vehicles consist of a pneumatic engine, an air-to-gaseous-nitrogen heat exchanger, and a cryogenic tank. The function of the tank is to provide both LN<sub>2</sub> fuel storage and primary evapora-



a)



b)



c)

**Fig. 1. Experimental cryogenic vehicles, operating on LN<sub>2</sub>:**

a) – UNT; b) – UW; c) – KU

tion. The maximum specific energy of nitrogen as a working fluid is estimated at 770 kJ/kg for a temperature difference between ambient air at 300 K and LN<sub>2</sub> at 77 K [1]. Using free thermal energy from the environment, the LN<sub>2</sub> is heated to release its stored energy and produce compressed gas to run the pneumatic engine. An advantage of LN<sub>2</sub> as an automobile fuel is the availability of abundant gaseous nitrogen in the atmosphere. When consumed for transportation, LN<sub>2</sub> is environmentally benign. Like hydrogen fuel, LN<sub>2</sub> is an energy carrier, not a source. So, energy must be invested to liquefy atmospheric nitrogen. However, this energy can be produced in a large stationary power plant with efficiency far exceeding internal combustion engines, and the effluent can be scrubbed or captured to mitigate pollution and greenhouse gasses. Moreover, the energy for liquefaction can also be obtained from alternative and non-polluting sources, such as solar and wind, which provide the opportunity to create self-contained “green” regions that produce and utilize LN<sub>2</sub> for pollution-free transport applications.

The critical constraint for cryogenic vehicles is fuel economy, which must be optimized to minimize the volume of LN<sub>2</sub> onboard. Special attention is given to achieving maximum efficiency in all parts of the cryogenic power system [7]. The typical function of a LN<sub>2</sub> Dewar is prolonged storage of cryogenic liquid by minimizing heat leak paths from the ambient into the liquid. For cryogenic vehicles, the storage tank serves two functions: 1) prolonged fuel storage when the vehicle is not in use (like a Dewar) and 2) rapid evaporation of the liquid to produce high-pressure gaseous nitrogen to drive the vehicle when it is in use [8]. From this later function arises the classical thermodynamic arrangement of heat moving along a temperature gradient from hot to cold; this configuration can be adapted for supplementary energy generation.

Miller, et al. suggest that thermoelectric generators topping an organic Rankin cycle will add to the overall energy of the system and thereby raise cycle efficiency [9].

Crane describes the potential for automotive high temperature waste heat recovery using thermoelectric generators in diesel vehicles [10]. Here, we evaluate placing a thermoelectric generator within the LN<sub>2</sub> storage tank wall of a cryogenic vehicle to recover part of the nitrogen’s latent heat of liquefaction as electricity. As we will demonstrate, this configuration presents a sufficient temperature gradient for useful thermoelectric generator applications. The recent creation of thermoelectric materials with high figure of merit at cryogenic temperatures further motivates analysis of this energy recovery technology for practical cryogenic automobile applications. Moreover, the presence of cryogenic liquid in a storage tank onboard the vehicle provides new synergistic opportunities. For example, generated electrical energy can be stored in the magnetic field of a high-temperature superconductor submerged in the tank, instead of as chemical energy stored in a battery as in conventional automobiles.

### Theory and Background

The thermoelectric material figure of merit,  $ZT$ , is a unit-less parameter defined as follows [11, 12]:

$$ZT = \frac{\alpha^2}{k\rho} \quad (1)$$

where  $\alpha$  is the Seebeck Coefficient [V/K],  $k$  is the thermal conductivity of the material [W/m<sup>2</sup>K], and  $\rho$  is the electrical resistivity of the material [Ohm·m].

Both p- and n-type semiconductor materials are used in making thermoelectric generators, and the properties of each are sometimes sufficiently different to warrant considering each material in estimating an overall figure of merit using the following equation [11,12]:

$$ZT = \frac{(\alpha_p - \alpha_n)^2}{[(k_p \rho_p)^{1/2} + (k_n \rho_n)^{1/2}]^2}, \quad (2)$$

where the subscripts p and n correspond to the p-type and n-type materials respectively.

These parameters arise from the geometry of p- and n-type legs of thermoelectric generators arranged in series, as shown in Fig. 2.

The efficiency of a thermoelectric generator is given by [11,12]

$$\eta = \frac{T_h - T_c}{T_h} \cdot \frac{M - 1}{M + \frac{T_c}{T_h}}, \quad (3)$$

where  $M$  is obtained by calculating

$$M = (1 + ZT_m)^{1/2} \quad (4)$$

and where  $T_m$  is the mean temperature of the material, derived from temperatures  $T_h$  and  $T_c$  of hot and cold ends of the thermoelectric material respectively.

A candidate thermoelectric material with high figure of merit at cryogenic temperatures has been identified: CsBi<sub>4</sub>Te<sub>6</sub> [13–15]. For our analysis, this material was selected on the basis of producing the most energy per unit mass of cryogen vaporized. The CsBi<sub>4</sub>Te<sub>6</sub> compound is a very recent discovery and can achieve as much as a 40% improvement in energy conversion efficiency over more traditional Bismuth Telluride compounds (see, for example, [14, 16]), which were previously the best cryogenic thermoelectric materials available. The temperature-dependent figures of merit for both materials are shown in Fig. 3 [14].

For estimation of the maximum performance of a thermoelectric generator placed within the walls of an automotive cryogenic tank, we used 300 K for  $T_h$  and 80 K for  $T_c$ . We also used a high value of  $ZT$  for CsBi<sub>4</sub>Te<sub>6</sub>, 0.75, which corresponds to the assembly's mean temperature value, 190 K. For these conditions, we calculate a maximum energy conversion efficiency of 14.9%. This value is the percentage of heat flowing into the generator's hot surface which gets converted to electricity. To estimate the maximum electricity generation, heat flow through the thermoelectric element was next calculated.

### Method for Estimating Heat Flow

The proposed thermoelectric assembly schematic, shown in Fig. 2, provides a basis for estimating heat transfer through the generator. The thermoelectric generator consists of two branches, one n-type and one t-type, which are selected to be 5 mm in length. Manufacturing limitations on conventional thermoelectric generators prevent that entire space from being filled with generator

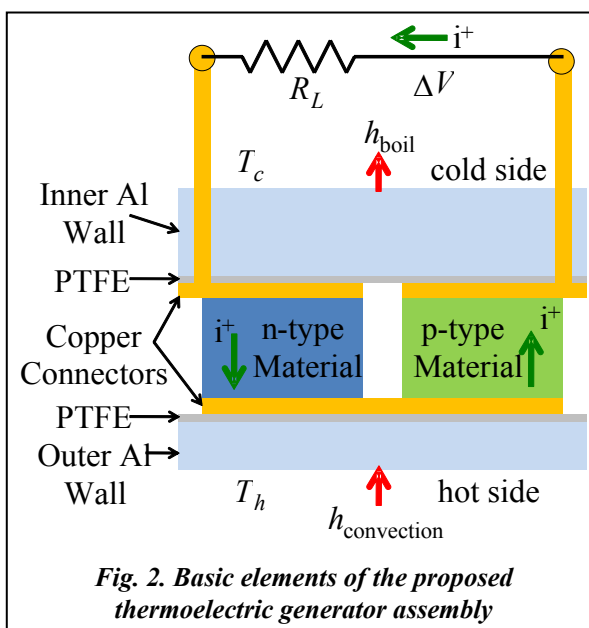


Fig. 2. Basic elements of the proposed thermoelectric generator assembly

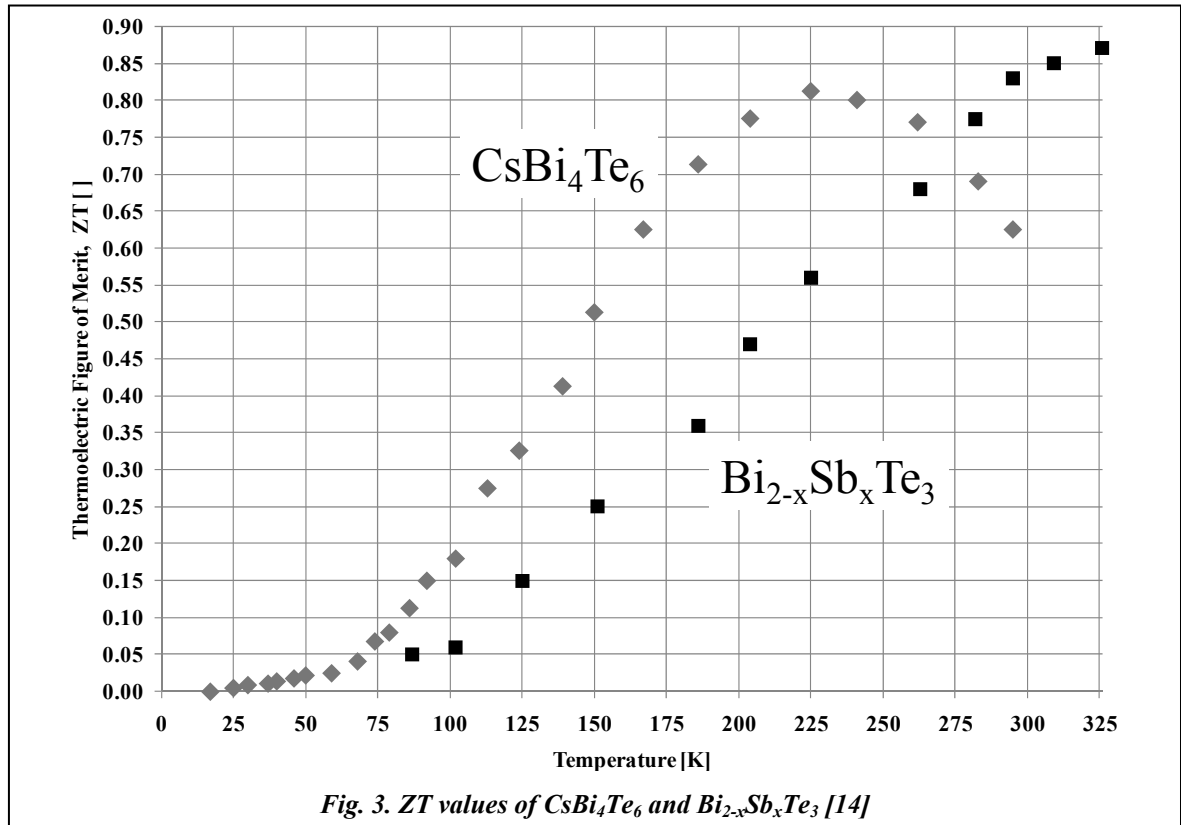


Fig. 3. ZT values of CsBi<sub>4</sub>Te<sub>6</sub> and Bi<sub>2-x</sub>Sb<sub>x</sub>Te<sub>3</sub> [14]

material, and a packing fraction (*PF*) of 0.4 to 0.6 is typical. We use *PF* = 0.5 for heat transfer calculations. The remaining open space is under vacuum, as in a Dewar, to restrict undesirable heat transfer around the thermoelectric generator. Heat transfer through this void space occurs by radiation, and we estimate its magnitude at less than 0.2% of the conduction through the thermoelectric legs. So, radiation is ignored.

The legs of thermoelectric material are sandwiched within two thin copper layers, which provide electrical contact. Electrical insulation between the copper and cryogenic tank walls is provided by layers of Polytetrafluoroethylene (PTFE), a material chosen for its mechanical and thermal stability at cryogenic temperature. PTFE lines the inner and outer walls of the cryogenic storage tank. The inner tank wall contacts stored LN<sub>2</sub> while the outer tank wall contacts the ambient environment.

Table 1. Properties of component materials

Material	Thickness	Thermal Conductivity
Units	m	W/(m K)
<b>Aluminum</b> [17]		
Inside (80 K)	0.006	432
Outside (300 K)	0.003	237
<b>Copper</b> [17]		
Inside (80 K)	0.001	557
Outside (300 K)	0.001	401
<b>Teflon</b> [18]		
Inside (80 K)	0.005	0.25
Outside (300 K)	0.005	0.28
<b>CsBi<sub>4</sub>Te<sub>6</sub></b> [14]	0.003	1.48

A one-dimensional heat transfer model was applied to the thermoelectric generator assembly to estimate maximum heat flux for a set of material properties and realistic assembly dimensions presented in Table 1. Cryogenic tank wall thicknesses were selected to withstand an internal pressure up to 3.5 MPa (500 psi). One-dimensional Cartesian heat transfer was assumed because the tank radius is large enough to neglect surface curvature local to the thermoelectric assembly.

For the cryogenic tank inner wall, which is in direct contact with liquid nitrogen, heat transfer to the liquid is assumed to occur via nucleate boiling. This assumption is justified be-

cause the calculated burn-out heat flux is about 135 times greater than the heat flux through the inner evaporator wall, and this wall is always at similar temperature to the liquid. The convective heat transfer coefficient is estimated from classic nucleate pool boiling correlation [19] on an upward-facing heated plate,

$$h_{\text{tank}} = \frac{k_l \text{Ja}^2}{C_{nb}^3 \text{Pr}_l^m \left[ \frac{\sigma}{(\rho_l - \rho_v)g} \right]^{1/2}} \quad (5)$$

where  $k_l$ ,  $\text{Ja}$ ,  $\text{Pr}$ ,  $\sigma$  are respectively the liquid's thermal conductivity, Jakob number, Prandlt number, and surface tension;  $\rho_l$  and  $\rho_v$  are respectively the liquid and gas densities; and  $g$  is the gravitational constant. The coefficient  $C_{nb}$  arises from experimental data and Mirza [20] suggests  $C_{nb} = 0.004$ , a value at the top of the range for smooth surfaces and the bottom of the range for rough surfaces. The exponent  $m$  also normally arises from experimental data, but an appropriate value could not be found in the literature for boiling nitrogen. Experimentally correlated  $m$  values range from 2 to 4.1 [21]. We therefore select  $m = 3$ , which is in the center of the range of available values for other boiling liquids. Given uncertainty in the exponent  $m$ , and noting that  $\text{Pr}$  for nitrogen in the cryogenic tank is about 2, Equation 5 predicts  $h_{\text{tank}}$  to an accuracy of  $\pm 54\%$ . The approach gives a convective heat transfer coefficient on the inner cryogenic tank wall of  $4890 \text{ W}/(\text{m}^2\text{K})$ , which compares favorably to  $8520 \text{ W}/(\text{m}^2\text{K})$ , a representative heat transfer coefficient calculated for boiling water at ambient pressure [22].

To estimate the natural convection heat transfer coefficient between the cryogenic tank's outside surface and ambient, a classical shape- and size-independent correlation [21] for a cooled plate facing downward is used

$$h_{\text{ambient}} = 0.14 \left( \frac{\rho_{\text{air}} \Delta\rho C_{p,\text{air}} g}{\mu_{\text{air}}} \right)^{1/3} k_{\text{air}}^{2/3} \quad (6)$$

where  $\rho_{\text{air}}$ ,  $C_{p,\text{air}}$ ,  $\mu_{\text{air}}$ , and  $k_{\text{air}}$  are the density, specific heat, viscosity and thermal conductivity respectively of air evaluated at ambient conditions (300 K and 1 atm), and  $\Delta\rho$  is the air density difference between ambient and the temperature of the tank's outside surface. Importantly, this approach assumes 1) natural convection only with no forced convection, in other word the tank and car are not moving; 2) no liquid condensate or ice build-up on the tank's outer surface; 3) the tank is elevated far enough off the ground that no obstructions interfere with the convection process; and 4) radiation heat transfer between the tank surface and the ground is negligible. Forced convection and radiation to the ground would tend to increase  $h_{\text{ambient}}$  while convection obstructions and ice build-up would tend to lower it. Equation 6 gives a convective heat transfer coefficient on the outer cryogenic tank wall of  $9.4 \text{ W}/(\text{m}^2\text{K})$ , which is within the generally accepted range of  $3\text{--}25 \text{ W}/(\text{m}^2\text{K})$  typical for this process [23].

## Results and Discussion

Table 2 gives thermal resistances for each element of the thermoelectric generator assembly. The greatest resistances to heat transfer occur at the outer wall, within the layers of PTFE insulation, and across the thermoelectric generator. Resistances in the metal layers are so comparatively small that they can be neglected.

The calculated energy conversion efficiency is 14.9 % and the heat flux entering the hot side of the thermoelectric generator is  $1477 \text{ W}/\text{m}^2$ . Therefore, the electrical energy generated per square meter of cryogenic storage tank is  $220 \text{ W}/\text{m}^2$ . The voltage/current condition can be adjusted, depending on the needs of the electrical system, by wiring multiple thermoelectric generator couples in parallel (to increase voltage) or in series (to increase current). Since heat absorption from the environment will take place only where the liquid is available for evaporation, it is not economical to line the entire cryogenic storage tank with thermoelectric generators. Instead, the generators should be placed at the bottom of the tank where gravity ensures liquid will always be present provided some fuel remains. Total added mass of the generators and total energy created are

**Table 2. Thermal Resistances of Elements in the Thermoelectric Generator Assembly**

<b>Outer Wall Convection</b>	<b>Outer Wall Conduction</b>	<b>Hot Side PTFE Conduction</b>
[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]
1.07E-01	1.27E-05	1.80E-02
<b>Hot Side Copper Conduction</b>	<b>Thermoelectric Generator</b>	<b>Cold Side Copper Conduction</b>
[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]
2.49E-06	1.01E-02	1.80E-06
<b>Cold Side PTFE Conduction</b>	<b>Inner Wall Conduction</b>	<b>Inner Wall Convection</b>
[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]	[m <sup>2</sup> -K/W]
1.99E-02	1.39E-05	2.05E-04

responding reduction in temperature gradient will drop the thermoelectric generator efficiency. The constrained temperatures used in this orientation model exaggerate the benefit of low PF and short thermoelectric generators. By reducing these two geometric parameters in this model, the total mass of thermoelectric assemblies in the wall of the cryogenic tank drops, but the efficiency is unaffected. Moreover, by reducing the generator length (thereby reducing the thermal resistance presented by the thermoelectric material), more heat flux is allowed through the generator, which appears to increase the total electrical work output because efficiency is unaffected. In reality, reducing the thermoelectric generator length also reduces the resistance to heat conduction through the thermoelectric generator assembly, which decreases the temperature gradient supported by the generator. Reduced temperature gradient drops the efficiency. We therefore expect these competing effects to yield an optimization problem resulting in thermoelectric generator lengths that give maximum power point, maximum volume-specific power, and maximum mass-specific power.

While this development will lead to an increase of the total efficiency of power system for cryogenic vehicles, the necessary presence of LN2 fuel motivates further areas of study in the judicious application of magnetic fields to enhance generator performance from the Ettinghausen effect [11]. On the other hand, it is well-known that thermoelectric generators produce high electric currents which can be applied for additional energy storage in superconducting magnetic energy storage (SMES) systems, based on high-temperature superconductors. These systems might be used on hybrid-electric cryogenic vehicles for propulsion [24, 25].

### Conclusion

Thus, due to appearance of new semiconductor thermo-elements that are effective at low temperatures it is possible to improve the design and efficiency of LN2 cryogenic evaporators by changing the first range of heat exchanger for a thermoelectric generator, to obtain additional electric power in amount of 14.9 % of heat flux, which was simply lost before. Using published parameters for CsBi<sub>4</sub>Te<sub>6</sub>, a thermoelectric material demonstrating good performance at low temperatures, we obtained a volume-specific power of 73·10<sup>4</sup> W/m<sup>3</sup> and a mass-specific power of 10.34 W/kg for cryogenic vehicle applications.

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therefore functions of each specific storage tank and were not calculated. However, volume-specific and mass-specific power can be calculated for comparison to established literature values. Dividing heat flux by thermoelectric length gives a volume-specific power of 73·10<sup>4</sup> W/m<sup>3</sup>, and dividing this power by the density of CsBi<sub>4</sub>Te<sub>6</sub> (7088 kg/m<sup>3</sup> [14]) gives a mass-specific power of 10.34 W/kg.

Importantly this performance analysis is optimistic because it fixes the temperatures on the hot and cold sides of the thermoelectric material at 300 K and 80 K respectively. In reality, these temperatures will each adjust, as governed by the thermal circuit made up of the thermoelectric assembly, to become closer to the mean assembly temperature. The cor-

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