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THE EFFECT OF HIGH-INTENSITY ELECTRON BEAM ON THE CRYSTAL STRUCTURE, PHASE COMPOSITION, AND PROPERTIES OF Al–Si ALLOYS WITH DIFFERENT SILICON CONTENT

The study deals with the element–phase composition, microstructure evolution, crystal-lattice parameter, and microdistortions as well as the size of the coherent scattering region in the Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu alloys irradiated with the high-intensity electron beam. As revealed by the methods of x-ray phase analysis, the principal phases in untreated alloys are the aluminium-based solid solution, silicon, intermetallics, and Fe₂Al₉Si₂ phase. In addition, the Cu₉Al₄ phase is detected in Al–10.65Si–2.11Cu alloy. Processing alloys with the pulsed electron beam induces the transformation of lattice parameters of Al–10.65Si–2.11Cu (aluminium-based solid solution) and Al–5.39Si–1.33Cu (Al₁ and Al₂ phases). The reason for the crystal-lattice parameter change in the Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu alloys is suggested to be the changing concentration of alloying elements in the solid solution of these phases. As established, if a density of electron beam is of 30 and 50 J/cm², the silicon and intermetallic compounds dissolve in the modified layer. The state-of-the-art methods of the physical materials science made possible to establish the formation of a layer with a nanocrystalline structure of the cell-type crystallization because of the material surface irradiation. The thickness of a modified layer depends on the parameters of the electron-beam treatment and reaches

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maximum of 90 μm at the energy density of 50 J/cm^2 . According to the transmission (TEM) and scanning (SEM) electron microscopy data, the silicon particles occupy the cell boundaries. Such changes in the structural and phase states of the materials response on their mechanical characteristics. To characterize the surface properties, the microhardness, wear parameter, and friction coefficient values are determined directly on the irradiated surface for all modification variants. As shown, the irradiation of the material surface with an intensive electron beam increases wear resistance and microhardness of the Al-10.65Si-2.11Cu and Al-5.39Si-1.33Cu alloys.

Keywords: cast aluminium alloys, electron beam treatment, microstructure, mechanical properties, structure and phase transformations, scanning electron microscopy, transmission electron microscopy, microhardness, tribological testing.

1. Introduction

The aluminium and silicon alloy is a commonly used material in motor vehicle and aerospace industries due to the excellent correlation between its strength and weight, outstanding castability, low density and sufficient corrosion resistance. Therefore, the Al-Si alloys are suitable for the manufacture of high-duty machine elements in motorcar and airplane industries, replacing traditional iron/steel [1–3]. However, Al-Si alloys suffer from certain limitations if compared with the steel, *e.g.* low strength and wear resistance characteristics of three-dimensional machine elements. It is the reason these alloys can hardly find a broad application. In view of these circumstances, different research groups have attempted to strengthen and modify these alloys as well as develop new technologies of their manufacturing [4, 5]. One of the main current discussions in this field is the effect of intensive plastic deformation on the structure and mechanical properties of alloys, *e.g.* equal channel angular pressing and equal channel angular extrusion [6–12]. Nevertheless, these methods are not suitable for the strengthening ready to use machine elements, so additional stages are needed in production process chains. In view of this, the diversity of processing techniques have been extensively developed, *e.g.* the microarc oxidation [13, 14], anodizing [15], plasma electrolytic oxidation [16, 17], plasma spraying [18], and electron beam treatment [19–25] — the surface processing methods, which adapt plasma, particles or electrochemical processes with the purpose to improve the wear resistance and corrosion properties of aluminium.

The surface treatment of metals and alloys with high current electron beams is recognized to be one of the promising and rapidly developing techniques. Within this technological process, accelerated electrons interact with a being processed surface, and as a result, their kinetic energy turns into the heat. A velocity of heating and cooling processes can be rather high (approximately $\sim 10^6$ K/s), cause therefore structure transformations and increase hardness. As a surface modification method, high-current electron-beam treatment is thought to have the advantage

over laser and pulsed ion processing techniques in terms of efficiency, simplicity, reliability and a beneficial effect on process properties of metallic materials [26–31]. Prior studies on a role of high current electron beams for the structure and physical and mechanical properties of aluminium alloys have pointed out this research domain is of high relevance. Independent groups of scientists have come to a conclusion high-current pulsed electron-beam treatment represents a general-purpose technique intended for the improving mechanical properties of Al–Si alloys. A number of studies have shown the element composition of the irradiation surface tends to change, thermal instabilities induced by rapid heating and cooling arise in material surface layers. This results into the evaporation and re-precipitation of some amount of alloying elements. A core component of the remelted surface layer is an over-saturated aluminium-based solid solution. The thickness of a remelted zone depends on the number of irradiation impulses. A number of experimental studies have recognized the better corrosion and mechanical properties of alloys treated with high current electron beams [32–38].

The goal of this paper is to explore the structure and phase states, crystal structure, and mechanical characteristics of the Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu alloys irradiated with the high current pulsed electron beam.

2. Material and Methods of Investigation

We used the cast aluminium Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu alloys as an experimental material. This compound is significant for the aluminium alloys of this chemical composition are used in the manufacture of a wide product assortment, *e.g.* household appliances, products of electrochemical and mechanical engineering. Parallelepiped samples with dimensions of $15 \times 15 \times 5$ mm³ were taken for processing and further research.

The chemical composition of as delivered samples determined with the use of x-ray spectrometry is given in Fig. 1. A surface of 15×15 mm² samples was polished and mirror-finished for the further electron-beam treatment. The electron beam was moved along the normal line to the polished side of a sample, a diameter of the beam was set to cover the area totally. The alloys were irradiated with the intensive pulsed electron beam using the ‘SOLO’ laboratory unit [39, 40]. The parameters of the electron beam were as follow: energy of accelerated electrons 17 keV; density of the electron beam 10, 30, and 50 J/cm²; pulse time 50 and 200 μ s; number of pulses 3; pulse repetition frequency 0.3 s⁻¹; and pressure of the residual gas (argon) in the processing chamber of the unit $2 \cdot 10^{-2}$ Pa. The phase composition was explored with the methods of x-ray phase analysis (Shimadzu x-ray diffractometer XRD 6000).

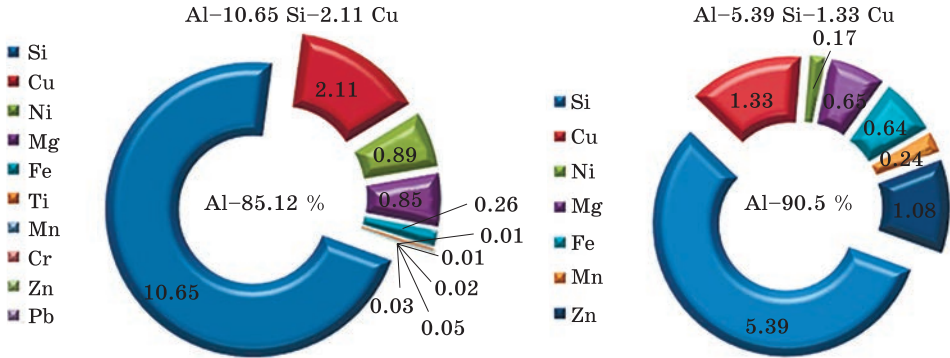


Fig. 1. The percentage of alloying elements and essential impurities in Al-10.65Si-2.11Cu and Al-5.39Si-1.33Cu alloys

The element composition and defect substructure state of alloys were examined using scanning electron microscopy (SEM) (Philips SEM-515 equipped with a microanalyser EDAX ECON IV) and transmission electron microscopy (TEM) (JEM 2100F to carry out high-sensitivity electron beam scanning (STEM analysis) and investigate the element composition of foils with the energy-dispersive analysis of x-rays). When carrying out a STEM analysis, the electron beam is focused into a 0.05–0.2 nm thin spot, further the beam scans a material of interest in the raster lighting system. STEM has the potential to visualize a nanostructured material surface with high spatial resolution and study the distribution of elements in the scanned zone. X-ray crystallography determined a structure state of the Al-10.65Si-2.11Cu and Al-5.39Si-1.33Cu alloys prior to and after the surface treatment with the intensive electron beam. With the use of this method, the lattice parameter and microdistortions as well as the size of the coherent scattering region were detected. X-ray patterns were produced with a scanning pitch of 0.05 degrees and an exposition time of 5 seconds in each point. Precision shooting of the highlighted x-ray peaks was performed with an interval of 0.02 degrees and an exposition time of 20 seconds. SEM was implemented directly on the treated surface and on etched cross-microsections of irradiated samples. Foils for exploring the structure and phase state of the material with the methods of transmission electron microscopy in the diffraction mode were prepared via ion beam sputtering of plates, which were cut perpendicular to the irradiated surface of a sample using the spark discharge method. A cutting mode was set precisely to avoid unnecessary deformation and therefore have no effect on the structure of a sample.

To characterize mechanical properties of the surface, microhardness testing was chosen in the study for it is one of the most precise and sensitive methods. Measurements were performed with a microhardness-

measuring device HVS-1000 using the Vickers method [41] and according to ISO 6507-1:2005 ‘Metallic materials. Vickers hardness test. Part 1. Test method’, restoring the indentation made by a four sided square-based pyramid. Loading was set constant and as high as $0.05HV$ for six processing modes. A time a load was applied and kept was 10 s, and within 5 s, a load was removed. Microhardness was determined directly on the modified surface and at different distances from it with a cross microsection.

Tribotechnical studies were implemented as ‘pin-on-disc’ testing (as specified in ASTM G99) using an Oscillating TRIBOtester (TRIBOtechnic) at load $P = 1$ N and sliding speed $V = 25$ mm/s. The diameter of a ShKh 15 steel counterbody was 6 mm, sliding distance $S = 20$ m, and wear track radius $r = 2$ mm.

3. Results and Discussion

3.1. The Element and Phase Composition, Crystal Structure of Al–Si Alloys Irradiated with the High-Current Pulsed Electron Beam

The element composition of the material surface was explored after electron beam irradiation. Figure 2 demonstrates the research findings of the Al–10.65Si–2.11Cu alloy element composition.

The elemental analysis of Al–10.65Si–2.11Cu alloy revealed the electron beam treatment (energy density 10 J/cm² and pulse time 50 μs) increases by 53.9% the silicon concentration in the material surface and its 63.4% growth was detected, if processed within 200 μs. A step-up in the beam energy density causes a 10–18% drop of the silicon concentration in the material surface in relation to process parameters. Another important fact is that the percentage of alloying elements can both fall and rise, depending on parameters of electron beam treatment.

The consideration of the effect of electron-beam treatment had on the silicon concentration in Al–5.39Si–1.33Cu alloy brought to light that the irradiation with a 10 J/cm² electron beam causes a 47.9–49.2% growth of a silicon share in the material, irrespectively to the pulse time. A percentage of silicon in the untreated Al–5.39Si–1.33Cu alloy is 5.39 wt.%, it increases up to 8.04 wt.% if irradiated with the electron beam (10 J/cm², 50 μs) and to 7.97 wt.% for a pulse time of 200 μs. The modification of the alloy surface with the pulsed electron beam heightens also the concentration of copper in the surface by $\approx 115.8\%$ and iron by $\approx 79.7\%$.

The outcomes of the elemental analysis of samples irradiated in different modes (Fig. 2 and 3) compared with the data on the chemical composition of untreated materials (Fig. 1) demonstrate that no matter what alloy grade is processed the electron beam density exerts a more

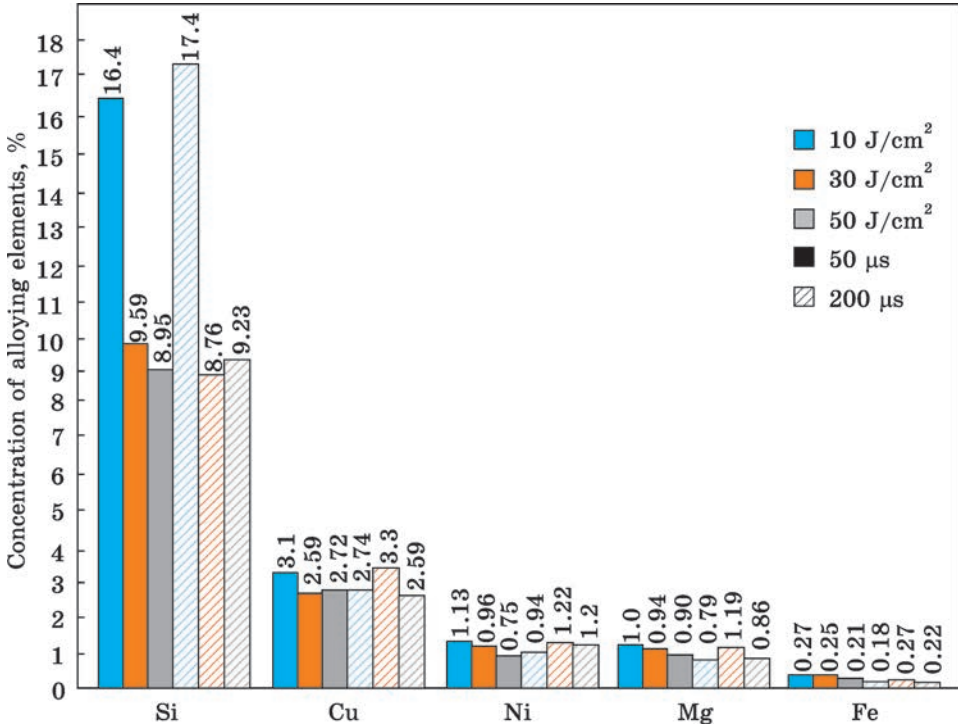


Fig. 2. The percentage of alloying elements and main impurities in Al-10.65Si-2.11Cu alloy irradiated with the pulsed electron beam of various energy density and pulse time

significant effect on the element composition of the modified layer than the beam pulse time. The results show the pulsed electron beam (energy density 10 J/cm²) plays the most important role for the silicon concentration in the material surface. This phenomenon is obviously associated with the fact that silicon is a more heat-resistant material than other basic chemical elements in the alloys, and the electron beam energy of 10 J/cm² is insufficient to melt it. An increase in the electron beam density up to 30 J/cm² and higher dissolves silicon and reduces, therefore, its concentration in the material.

Studies on the phase composition of Al-5.39Si-1.33Cu and Al-10.65Si-2.11Cu alloys identified that principal phases in the cast materials of interest include an aluminium-based solid solution, silicon and intermetallic compounds, e.g., a Fe₂Al₉Si₂ phase. In the Al-10.65Si-2.11Cu alloy, there are also Cu₉Al₄, Cu_{8.92}Al_{4.08}, Cu_{5.64}Al_{4.61} phases undetected in the Al-5.39Si-1.33Cu alloy.

An x-ray diffraction pattern of the Al-5.39Si-1.33Cu alloy produced in the x-ray phase analysis is provided in Fig. 4. The diffraction maximums of aluminium for the Al-5.39Si-1.33Cu alloy show their

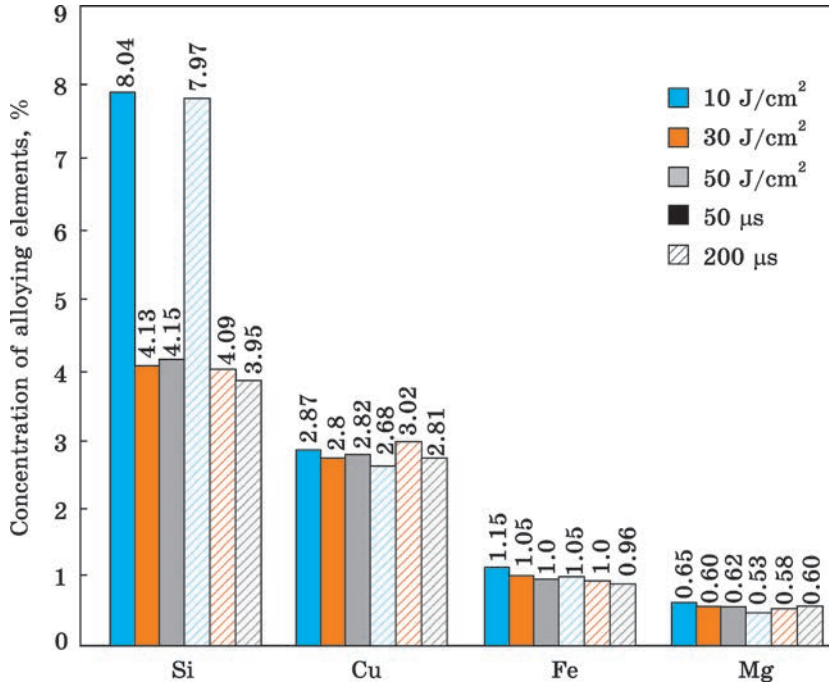


Fig. 3. The percentage of alloying elements and main impurities in Al-5.39Si-1.33Cu alloy irradiated with the pulsed electron beam of various energy density and pulse time

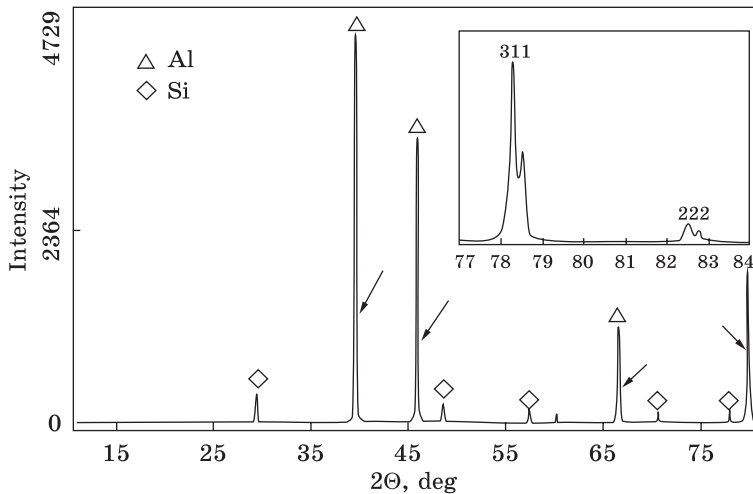


Fig. 4. A fragment of the x-ray diffraction pattern for the untreated Al-5.39Si-1.33Cu alloy. Arrows indicate the position of Al₂ diffraction lines

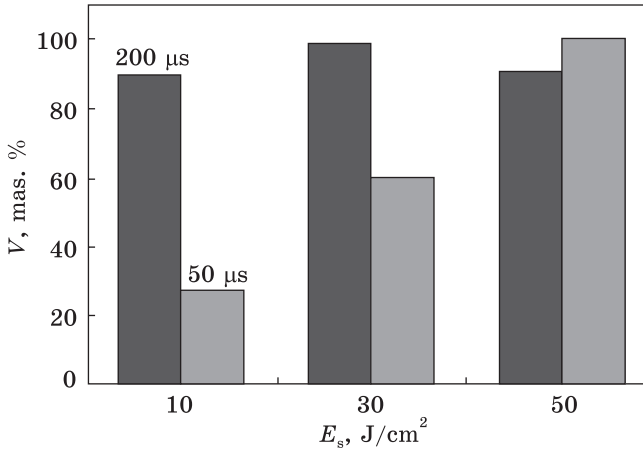


Fig. 5. The percentage of the Al₂ phase in the surface layer of Al-5.39Si-1.33Cu alloy vs. the energy density of electron beam

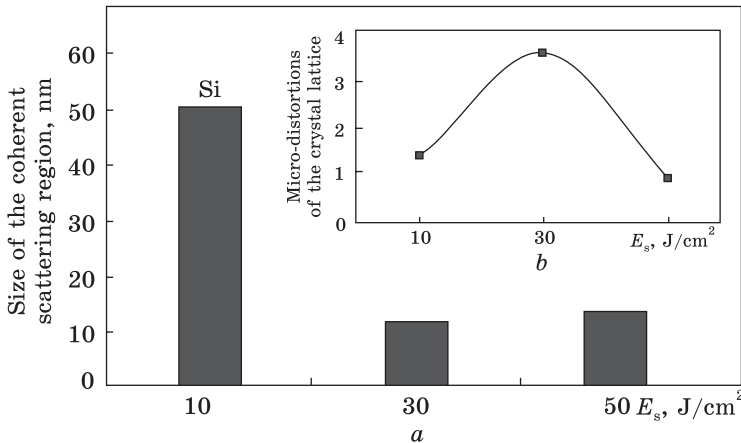


Fig. 6. The size of the coherent scattering region (a) and micro-distortions of the crystal lattice (b) vs. the energy density of the electron beam in Al-5.39Si-1.33Cu alloy

splitting up (the element added to Fig. 4). These data assume there are two aluminium-based solid solutions with a different lattice parameter. The first type Al₁ is an aluminium-based solid solution with a bigger lattice parameter; the second one Al₂ has a smaller lattice parameter. Some findings emerging from these data relate to the percentages of Al₁ (75.1 mas.%), Al₂ (20.0 mas.%), and silicon (the rest) in the Al-5.39Si-1.33Cu alloy prior to electron beam irradiation.

Existing research found the relative percentage of the Al₂ phase increases if the Al-5.39Si-1.33Cu alloy is treated with the pulsed electron beam (pulse time 50 μ s) and its energy density is raised (Fig. 5). The percentage of the Al₂ phase attains its maximum of \approx 99 mas.% for a pulse time of 200 μ s and the electron beam energy density of 30 J/cm² (Fig. 5).

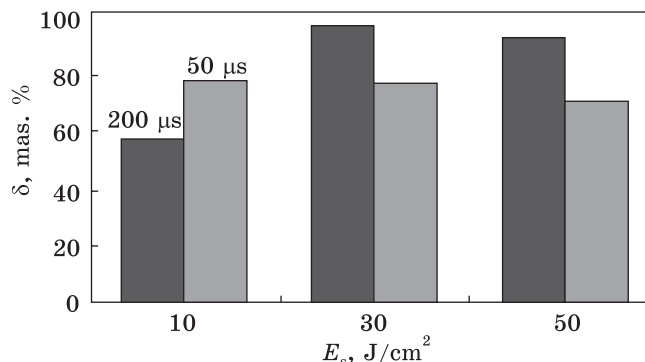


Fig. 7. The percentage of an aluminium-based solid solution in the surface of Al–10.65Si–2.11Cu alloy vs. the energy density of the electron beam

Studies on the size of the coherent scattering region of Al–5.39Si–1.33Cu alloy electrons established it extends to 51.02 nm for the Si phase given the electron-beam energy density is 10 J/cm² and pulse time is 200 μs (Fig. 6, a). A raise of the beam energy density up to 30 J/cm² results in the shrinkage of the coherent scattering region to 12.08 nm, whereas it is 1.99 nm for the beam energy density of 50 J/cm². From the data in Fig. 6, b, it is apparent the effect is contrary for microdistortions of the Si crystal lattice; that is, they are $1.396 \cdot 10^{-3}$ at 10 J/cm², rise up to $3.576 \cdot 10^{-3}$ at 30 J/cm² and drop to $0.934 \cdot 10^{-3}$ at 50 J/cm².

As observed, the irradiation of the Al–10.65Si–2.11Cu alloy with the pulsed electron beam changes the phase composition of the surface. To be more precise, a percentage of an aluminium-based solid solution increases in samples treated with the electron beam (a pulse time of 200 μs) (Fig. 7) and drops in samples irradiated with the electron beam (a pulse time of 50 μs) (Fig. 7) given the density of the electron beam is heightened.

Simultaneously, the content and element composition of hardening phases changes in the surface of the irradiated Al–10.65Si–2.11Cu alloy. The data (Fig. 8) demonstrate the silicon percentage decreases and attains its minimum at the energy density of 30 J/cm² provided that a pulse time of the electron beam is set 200 μs and the beam energy density grows (Fig. 8).

The sum total percentage of intermetallic compounds (Cu_9Al_4 , $\text{Cu}_{8.92}\text{Al}_{4.08}$, $\text{Cu}_{5.64}\text{Al}_{4.61}$) also shows a decreasing trend, becoming zero at the energy density of 50 J/cm² and a pulse time of the electron beam of 200 μs (Fig. 8, curve 3). Once a pulse time of the electron beam is 50 μs, there are opposite changes in the phase composition of the surface. To illustrate, an increase of the beam energy density has an insignificant effect on the silicon percentage (Fig. 8, curve 2). The sum total percentage of intermetallic compounds ($\text{Cu}_{8.92}\text{Al}_{4.08}$, $\text{Cu}_7\text{Si}_{12}$) increases and becomes maximal at the energy density of 50 J/cm² (Fig. 8, curve 4).

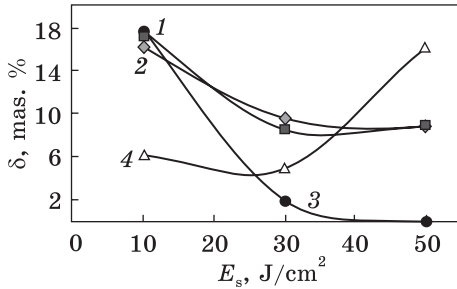


Fig. 8. The role of the beam energy density for the percentage of silicon (curve 1 and 2) and intermetallic compounds (curve 3 and 4) in the surface of Al-10.65Si-2.11Cu alloy irradiated with the pulsed electron beam; 1 and 3 — pulse time 200 μ s; 2 and 4 — 50 μ s

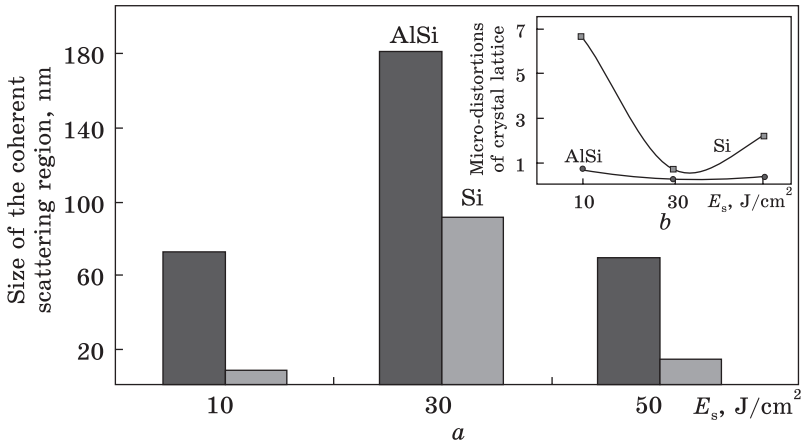


Fig. 9. The size of the coherent scattering region (a) and lattice microdistortions (b) vs. the energy density of electron beam for Al-10.65Si-2.11Cu

Sizes of the coherent scattering region for the Al-10.65Si-2.11Cu alloy were found for two phases — AlSi and Si. The study established the function connecting a size of the coherent scattering region and energy density of the electron beam is a curve with the maximum at 30 J/cm² irrespectively to the selected phase (Fig. 9, a). However, the maximal coherent scattering region for Si is 92.92 nm; that is almost twice as small as the maximal coherent scattering region for the AlSi phase (183.42 nm). The function of lattice microdistortions vs. energy density of the electron beam, on the contrary, is a curve with the minimum observed similarly to the energy density of the electron beam of 30 J/cm²; it was determined to be $0.744 \cdot 10^{-3}$ for Si and $0.287 \cdot 10^{-3}$ for AlSi (Fig. 10, b).

The changing phase composition and percentage of phases when irradiating Al-5.39Si-1.33Cu and Al-10.65Si-2.11Cu alloys with the electron beam are related to transformations in the lattice parameter of the main phase (Al).

Since two phases of aluminium-based solid solutions with different lattice parameters were detected in Al-5.39Si-1.33Cu, each phase was

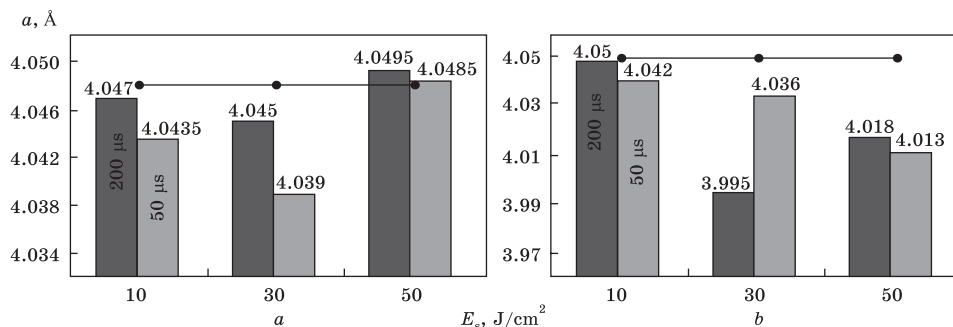


Fig. 10. The lattice parameter of the Al₁ phase (a) and Al₂ phase (b) forming in the surface layer of Al–5.39Si–1.33Cu alloy vs. the energy density of electron beam. The unbroken black line indicates the lattice parameter in the untreated alloy

analysed (Fig. 10). The data presented in Fig. 10, a demonstrate the lattice parameter of Al₁ is 4.047 Å at the beam energy density of 10 J/cm² and a pulse time of 200 μs, and 4.0435 Å at a pulse time of 50 μs, respectively. The lattice parameter of the Al₁ phase varies as indicated in the graph (Fig. 10, a) and its minimal values are at the beam energy density of 30 J/cm² (4.039 Å for 50 μs; 4.045 Å for 200 μs). At the beam energy density of 50 J/cm², the lattice parameter of the Al₁ phase (4.0485 Å for 50 μs; 4.0495 Å for 200 μs) is higher than the lattice parameter of the Al₁ phase in the untreated material (4.048 Å).

From the data in Fig. 10, b, it is apparent that the lattice parameter of Al₂ phase (4.05 Å) is similar to this characteristic of the Al₂ phase in the untreated material at the beam energy density of 10 J/cm² and a pulse time of 200 μs. As a pulse time is set to 50 μs, the lattice parameter of the Al₂ phase (4.042 Å) is lower than this characteristic of the Al₂ phase in the untreated alloy. The lattice parameter of the Al₂ phase at a pulse time 200 μs shows a similar trend as the lattice parameter of the Al₁ phase, attaining its lowest value at the beam energy density of 30 J/cm² (3.995 Å). Once a pulse time of the electron beam is 50 μs, the lattice parameter of the Al₂ phase decreases under the rising energy density of the electron beam and becomes minimal (4.013 Å) at 50 J/cm².

The behaviour of the lattice parameter of the aluminium-based solid solution and its dependence on the energy density of the electron beam in the Al–10.65Si–2.11Cu alloy is presented in Fig. 11. As seen in Fig. 11, the lattice parameter of aluminium changes irrespectively to the pulse time, its minimums are detected at a pulse time of 200 μs (4.0382 Å) for the energy density of 10 J/cm², and at a pulse time of 50 μs (4.0358 Å) for the energy density of 30 J/cm².

It is obvious that the principal reason for the mentioned lattice parameter changes of Al phases no matter which alloy (Al–5.39Si–1.33Cu

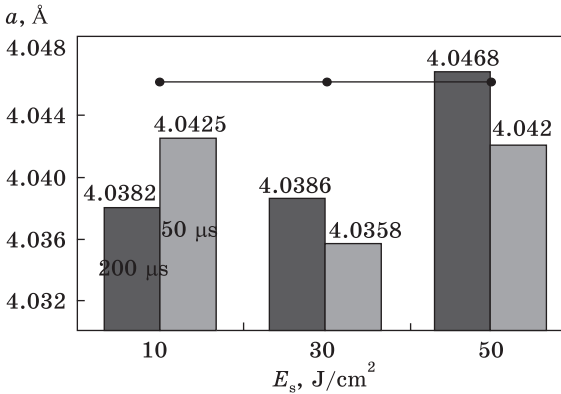


Fig. 11. The lattice parameter of an aluminium-based solid solution forming in the surface of the Al-10.65Si-2.11Cu alloy vs. the energy density of electron beam. The solid black line shows the lattice parameter of the phase in the untreated alloy

or Al-10.65Si-2.11Cu) is under study is the varying concentration of alloying elements. As known, the radii of silicon, copper, nickel, iron and manganese atoms are smaller, whereas the magnesium radius is bigger than the radius of the aluminium atom [42]. Therefore, lattice parameters of Al phases depend on the percentage of these elements in the solid solution. The relation between the lattice parameter of Al phases and pulse time, energy density of the electron beam is attributed to the dissolution processes of silicon particles and intermetallic compounds, as well as their re-precipitation occurring if a material is irradiated with the pulsed electron beam.

3.2. Structure Evolution of Al-Si Alloys Irradiated with the High Current Pulsed Electron Beam

3.2.1. Studying the Structure of Al-5.39Si-1.33Cu and Al-10.65Si-2.11Cu by the SEM Methods

The SEM-produced images of the untreated material structure are presented in Fig. 12. Dimensionally, aluminium grains are in a range from 25 μm to 100 μm , and the sizes of Al-Si eutectics grains varies from 10 to 30 μm for the Al-10.65Si-2.11Cu alloy. In the Al-5.39Si-1.33Cu alloy, an aluminium grain ranges from 25 μm to 80 μm , whereas the size of Al-Si eutectics grains is determined to be 11-26 μm , i.e., the structure elements in both as cast alloys have similar sizes. It is important that there are particles of intermetallic compounds in the alloys, which cause the brittleness of materials under loads.

The SEM of the modified alloy surfaces revealed no significant structural transformations in the Al-5.39Si-1.33Cu alloy at the energy density of 10 J/cm^2 , and pulse time settings (50 μs or 200 μs) were not critical (Fig. 13, a and d) if compared to the untreated alloy structure (Fig. 12, a). Micropores and second phase inclusions of a variety of forms and sizes were detected.

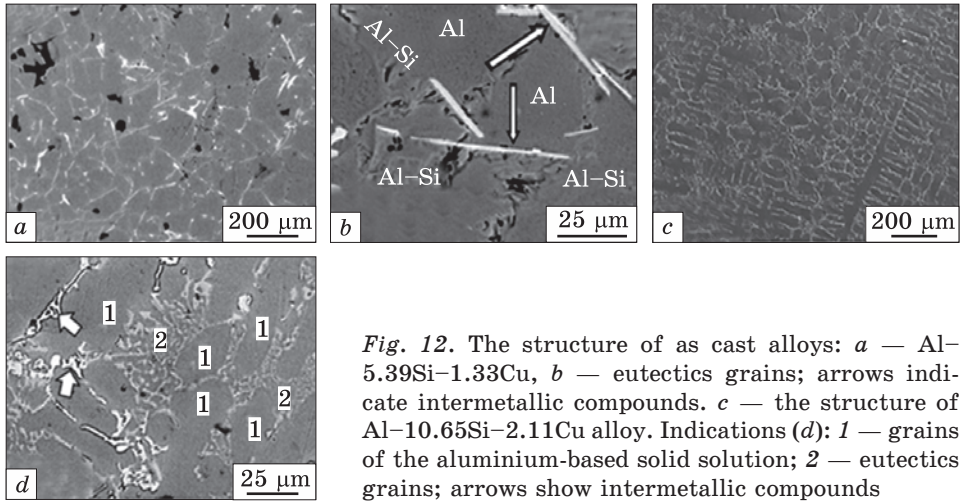


Fig. 12. The structure of as cast alloys: *a* — Al-5.39Si-1.33Cu, *b* — eutectics grains; arrows indicate intermetallic compounds. *c* — the structure of Al-10.65Si-2.11Cu alloy. Indications (*d*): 1 — grains of the aluminium-based solid solution; 2 — eutectics grains; arrows show intermetallic compounds

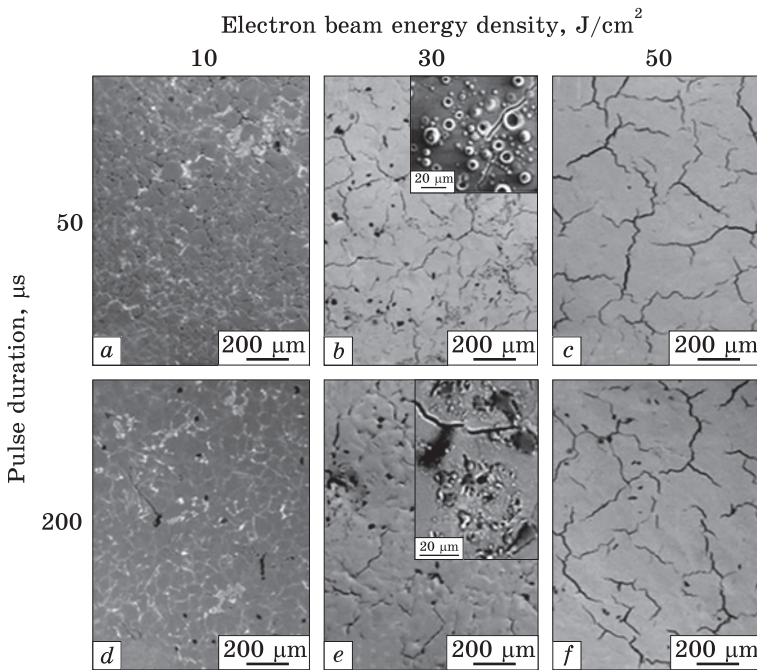


Fig. 13. The surface structure of the Al-5.39Si-1.33Cu alloy irradiated with the pulsed electron beam in various modes [43]

When the beam energy density was increased to 30 and 50 J/cm², the surface layer experiences essential transformations, *i.e.*, the melting and dissolution of second phase particles, the development of hardening cracks along the grain boundaries. If the energy density of the electron

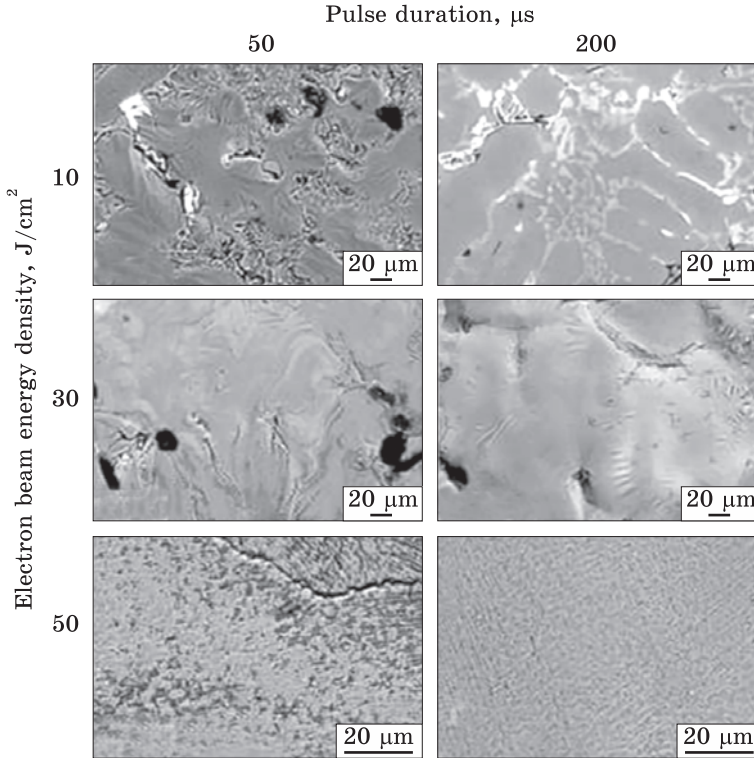


Fig. 14. The structure of Al-10.65Si-2.11Cu alloy surface irradiated with the pulsed electron beam in different regimes [44]

beam is of 30 J/cm^2 , second phase particles (intermetallic compounds and silicon) fail to dissolve totally. On the irradiated surface, there are regions with microcraters and globular inclusions (see insets in Fig. 13, *b* and *e*). It is interesting that a number of zones with partially dissolved second phase particles and microcraters is higher in the surface layer of samples irradiated with the electron beam of a pulse time of $50 \mu\text{s}$ than in the surface of the Al-5.39Si-1.33Cu alloy after irradiating with the electron beam of a pulse time of $200 \mu\text{s}$. The pulse time has virtually no effect on the irradiated surface state at the energy density of the electron beam of 50 J/cm^2 . On the treated surface, there are microcracks splitting the surface into regions of tens or hundreds of microns (Fig. 13, *c*, *f*). Scanning electron microscopy found no second phase particles (intermetallic compounds and silicon).

The SEM of the Al-10.65Si-2.11Cu alloy surface irradiated with different values of energy density and pulse time (Fig. 14) shows the treated surface becomes wavy at the beam energy density of 10 J/cm^2 ; that evidences the beginning of its melting. Similarly to the as cast

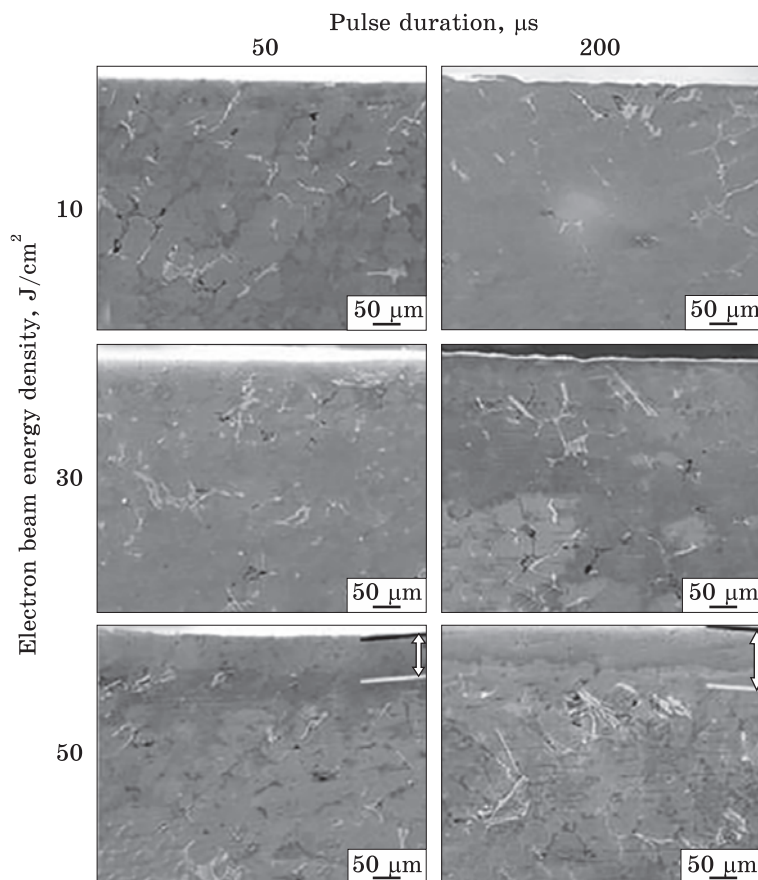


Fig. 15. The structure of the etched Al-10.65Si-2.11Cu micro-section irradiated with the pulsed electron beam in different regimes [45]

state, the structure has three phases: grains of an aluminium-based solid solution, eutectics grains and intermetallic compounds. At the energy density of 30 J/cm^2 , intermetallic compounds dissolve and numerous micropores appear that is probably the consequence of the material shrinkage caused by its rapid solidification. Chaotically located micropores are detected on the irradiated surface at the beam energy density of 50 J/cm^2 irrespectively to the pulse time; they are formed because of inner stresses in the material surface induced by the rapid solidification of the molten layer.

To explore the Al-10.65Si-2.11Cu alloy, the SEM was performed on microsections perpendicular to the irradiated surface (Fig. 15).

The data suggest that the thickness of the modified layer correlates with the beam energy density, *e.g.*, 10 J/cm^2 — $1\text{--}2 \text{ }\mu\text{m}$; 30 J/cm^2 —

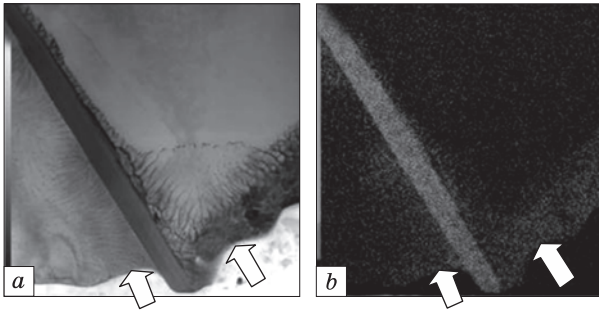


Fig. 16. TEM data of the surface in Al-5.39Si-1.33Cu alloy irradiated with the pulsed electron beam (10 J/cm^2 , $50 \text{ }\mu\text{s}$, 3 pulses) (a) and a view of this layer produced in the characteristic x-ray radiation of silicon atoms (b). Arrows indicate the irradiation surface

$25\text{--}30 \text{ }\mu\text{m}$; 50 J/cm^2 — $60\text{--}90 \text{ }\mu\text{m}$. The effect of the electron beam of 30 and 50 J/cm^2 is the dissolution of silicon and intermetallic compounds found in the untreated material; the pulse time is unimportant.

3.2.2. TEM Analysis of Thin Structure in Al-5.39Si-1.33Cu

Studies included also the assessment of the thin structure in alloys with the methods of transmission electron microscopy (TEM) before and after irradiating with the pulsed electron beam. Research results show the electron beam modification of the Al-5.39Si-1.33Cu alloy with the fixed parameters, 10 J/cm^2 ($50 \text{ }\mu\text{s}$, 3 pulses), result in a dendrite- and cell-type structure of rapid solidification around silicon lamellae (Fig. 16).

Electron microdiffraction patterns (Fig. 17, b) and dark-field images (Fig. 17, c, d) show the crystallization cells formed by the aluminium-based solid solution; on the cell boundaries of aluminium, there are particles of silicon.

The crystallization cells are in a range from 40 nm to 100 nm ; the sizes of silicon particles are $5\text{--}10 \text{ nm}$. The thin layers of silicon, which form the structure of lamellar eutectics (Fig. 17, a, d) vary from 15 to 25 nm . Studies established FeSi_2 phase lamellae split into several 200 to 300 nm roundish (globular) fragments in the surface irradiated with the electron beam (Fig. 18).

An increase of the beam energy density up to $30\text{--}50 \text{ J/cm}^2$ with no respect to the pulse time ($50 \text{ }\mu\text{s}$ or $200 \text{ }\mu\text{s}$) results in a rapid crystallization structure in a $50\text{--}70 \text{ }\mu\text{m}$ layer. The x-ray spectroscopy shows cells are built by an aluminium-based solid solution; the layers separating crystallization cells contain atoms of silicon, copper, and iron (Fig. 19).

The element composition of a foil section presented in Fig. 19, a is as follows: magnesium ($0.85 \text{ mas.}\%$), silicon ($2.83 \text{ mas.}\%$), manganese ($0.25 \text{ mas.}\%$), iron ($0.54 \text{ mas.}\%$), nickel ($0.06 \text{ mas.}\%$), and copper ($4.07 \text{ mas.}\%$), balanced by aluminium.

A characteristic view of the cell-type crystallization structure produced with the TEM methods is presented in Fig. 20. A dark-field anal-

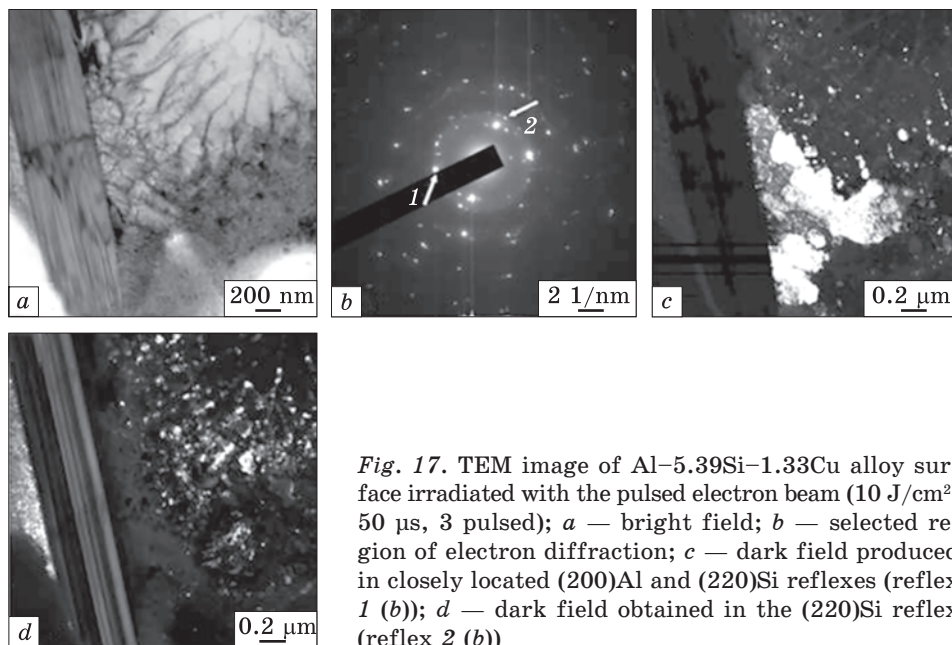


Fig. 17. TEM image of Al–5.39Si–1.33Cu alloy surface irradiated with the pulsed electron beam (10 J/cm^2 , $50 \text{ }\mu\text{s}$, 3 pulsed); *a* — bright field; *b* — selected region of electron diffraction; *c* — dark field produced in closely located (200)Al and (220)Si reflexes (reflex 1 (*b*)); *d* — dark field obtained in the (220)Si reflex (reflex 2 (*b*))

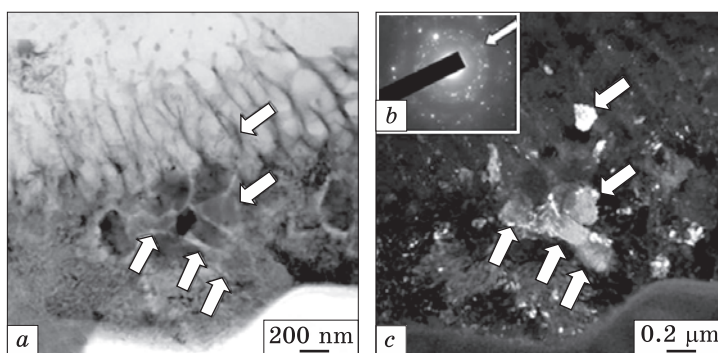


Fig. 18. TEM image of Al–5.39Si–1.33Cu alloy surface irradiated with the pulsed electron beam (10 J/cm^2 , $50 \text{ }\mu\text{s}$, 3 pulses); *a* — bright field; *b* — selected region of electron diffraction; *c* — dark field produced in the (102)FeSi₂ reflex (indicated with arrow (*b*)); arrows (*a*, *c*) indicate FeSi₂ particles

ysis highlights the volume of cells consists of an aluminium-based solid solution (Fig. 20, *c*). The cells are separated by second-phase particles with 10–20 nm crystallites (Fig. 20, *d*). A diffraction microanalysis suggests a main element in these crystallites is silicon, however, there can be also complex-composition particles containing copper, iron, silicon, and aluminium atoms.

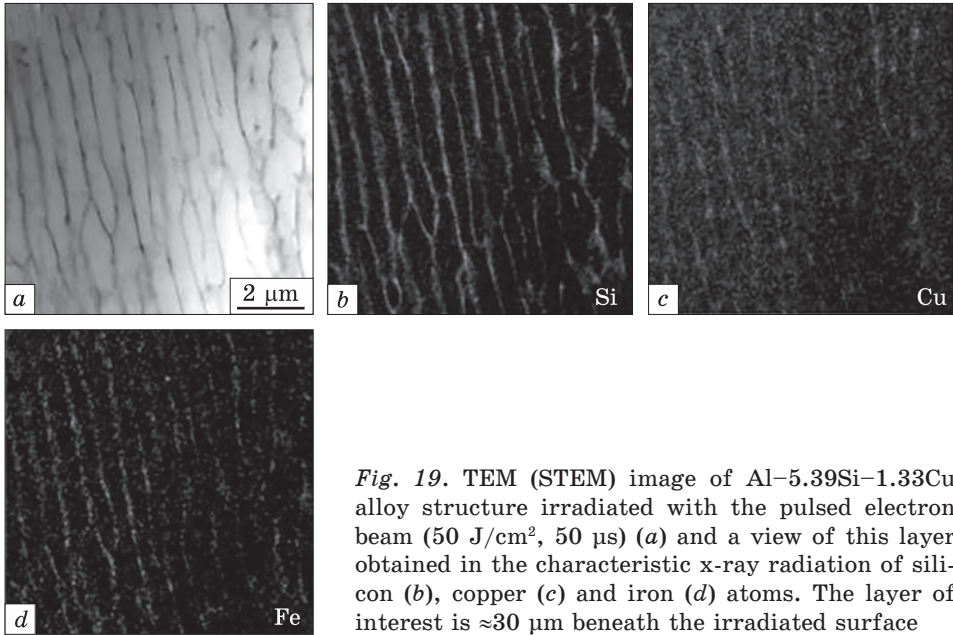


Fig. 19. TEM (STEM) image of Al-5.39Si-1.33Cu alloy structure irradiated with the pulsed electron beam (50 J/cm^2 , $50 \mu\text{s}$) (a) and a view of this layer obtained in the characteristic x-ray radiation of silicon (b), copper (c) and iron (d) atoms. The layer of interest is $\approx 30 \mu\text{m}$ beneath the irradiated surface

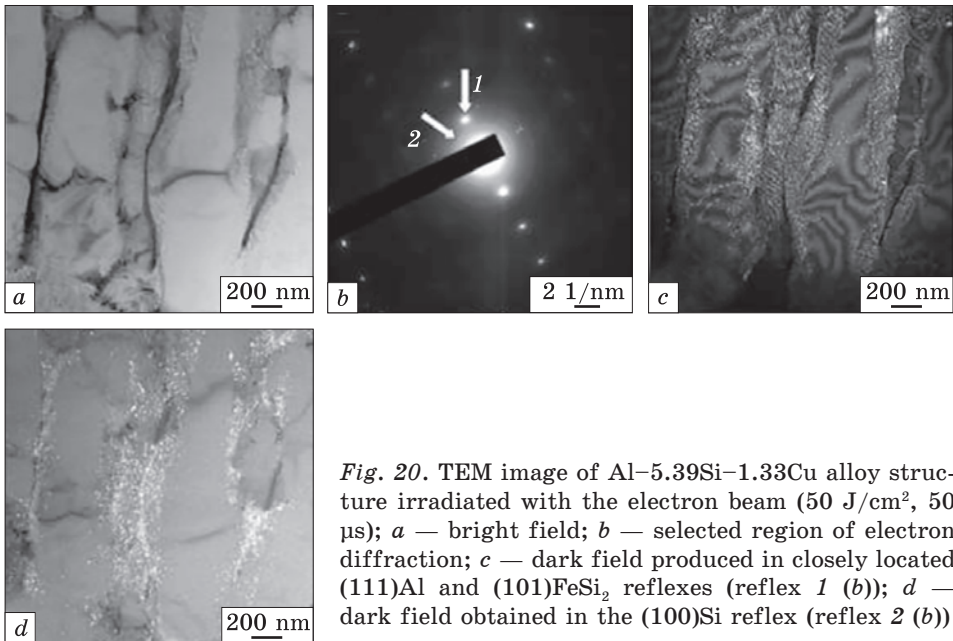


Fig. 20. TEM image of Al-5.39Si-1.33Cu alloy structure irradiated with the electron beam (50 J/cm^2 , $50 \mu\text{s}$); a — bright field; b — selected region of electron diffraction; c — dark field produced in closely located (111)Al and (101)FeSi₂ reflexes (reflex 1 (b)); d — dark field obtained in the (100)Si reflex (reflex 2 (b))

3.2.3. Studying the Thin Structure in the Al–10.65Si–2.11Cu Exposed to the High Current Pulsed Electron Beam

A TEM analysis of the Al–10.65Si–2.11Cu alloy demonstrates the outcome of the electron beam treatment with the energy density of 10 J/cm^2 (irrespective to the pulse time of the electron beam) is the melting of aluminium in the most zones of the processed surface. Crystallites of silicon and intermetallic compounds tend to dissolve partially because the electron beam has a weak thermal impact. The STEM studies disclosed a structure forming due to the unfinished dissolution of a silicon crystallite (Fig. 21).

From the data, it is apparent an aluminium structure of the rapid cell-type crystallization is formed after high-velocity heating and cooling (Fig. 21, *a*). The cell boundaries are surrounded with second phase thin layers containing mostly silicon atoms (Fig. 21, *b*). Of particular importance is that the thickness of the modified layer surrounding silicon crystallites reduces with distance from the irradiated surface and this layer disappears completely at a depth of 10–12 μm . This fact points indirectly at the thickness of a layer where aluminium melts in the contact zone with silicon crystallites.

The outcomes of the electron microscopy diffraction microanalysis carried out on a foil section shown in oval (Fig. 21, *a*) are given in Fig. 22. As seen from the data in Fig. 22, *a*, the aluminium cells of the rapid crystallization are oval; cross sizes of cells are in a range from 100 nm to 300 nm and lengthwise the cells are from 150 nm to 450 nm. The 40–75 nm thin second-phase layers surround the cells. The tin layers have a nanosize structure; their crystallites vary from 5 nm to 10 nm (Fig. 22, *c*). The electron diffraction micropattern obtained on this foil section (Fig. 22, *b*) shows these thin layers are composed of silicon par-

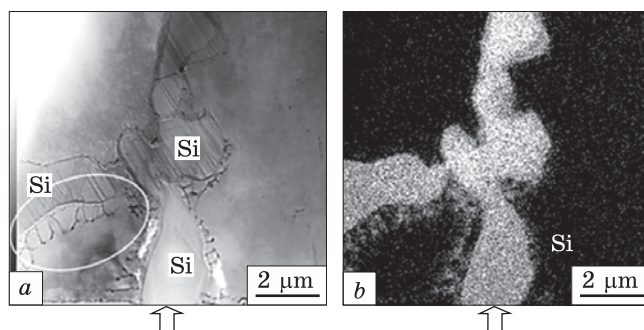


Fig. 21. STEM image (*a*) and an image produced in the characteristic x-ray radiation of silicon atoms (*b*) of Al–10.65Si–2.11Cu alloy surface irradiated with the pulsed electron beam at the beam energy density of 10 J/cm^2 , 200 μs . The arrows indicate the sample surface exposed to irradiation. The oval (*a*) shows a foil region explored in the electron microscopy diffraction microanalysis [45]

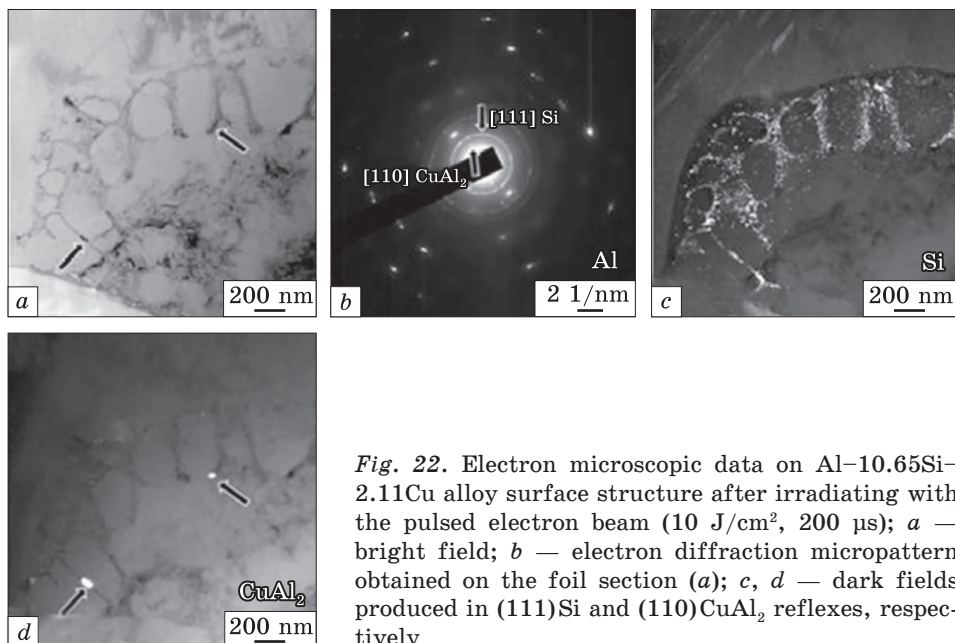


Fig. 22. Electron microscopic data on Al-10.65Si-2.11Cu alloy surface structure after irradiating with the pulsed electron beam (10 J/cm², 200 μs); *a* — bright field; *b* — electron diffraction micropattern obtained on the foil section (*a*); *c*, *d* — dark fields produced in (111)Si and (110)CuAl₂ reflexes, respectively

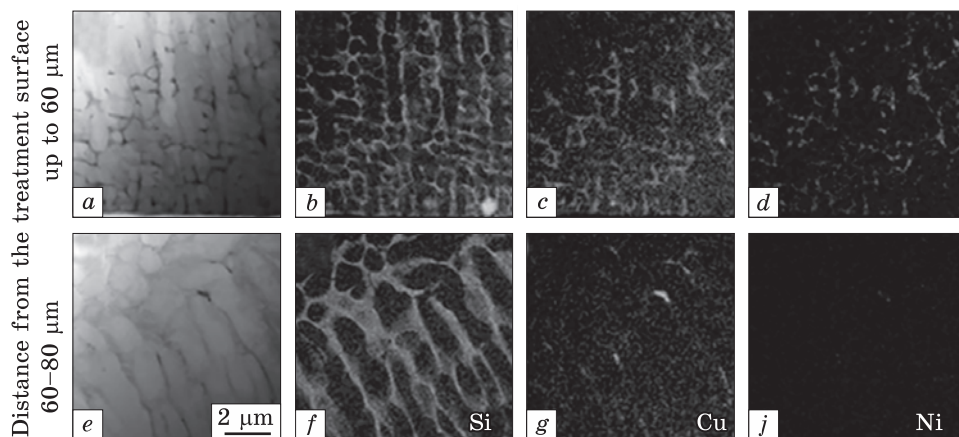


Fig. 23. STEM views (*a*, *e*) and data obtained in the characteristic x-ray radiation of silicon (*b*, *f*), copper (*c*, *g*) and nickel (*d*, *j*) atoms of Al-10.65Si-2.11Cu alloy surface irradiated with the pulsed electron beam at the beam energy density of 50 J/cm²; *a-d* — the surface not deeper than 60 μm; *e-j* — the layer at the depth of 60–80 μm [46]

ticles (Fig. 22, *c*). Nanosize particles of silicon are detected also within the volume of cells. Besides silicon particles, other phases can form fragments in thin layers. Figure 22, *d* summarizes the results indicating the CuAl₂ phase forming in thin layers.

The thickness of a layer containing the rapid cell-type crystallization structure increases with the growing energy density of the electron beam (a pulse time is unimportant), and its value is of $\approx 90 \mu\text{m}$ for 50 J/cm^2 . The x-ray spectroscopy found out the element composition of thin layers surrounding crystallization cells is connected with the distance to the irradiated surface. In the $60 \mu\text{m}$ surface in thin layers between crystallization cells, there are atoms of silicon, copper, nickel, and an insignificant amount of iron and magnesium (Fig. 23, *a–d*). Taking into account the element composition of thin layers, we assume that this layer of the alloy is a result of melting and rapid solidification of all phases constituting the material (aluminium, silicon, intermetallic compounds).

There are atoms of silicon and a small amount of copper $60\text{--}80 \mu\text{m}$ below the irradiation surface in thin layers surrounding the rapid solidification cells (Fig. 23, *e–j*). This layer of the material is suggested to be a product of melting and further rapid solidification of aluminium grains and silicon crystals. Fragments of intermetallic compounds hardly dissolved in this layer.

A structure with silicon crystallites, a cell-type crystallization structure around them and intermetallic compounds are detected $80\text{--}90 \mu\text{m}$ under the irradiation surface. Apparently, the conditions in this layer promote the melting of aluminium and partial dissolution of silicon crystallites. The further rapid solidification resulted in a cell-type crystallization structure in the material close to silicon crystallites. A structure similar to that of the untreated material is found with distance from the irradiated surface.

To sum up, the surface of the Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu alloys modified with the intensive pulsed electron beam has a cell-type crystallization structure with a variety of cell sizes. The SEM and TEM studies give the information on silicon particles on the cell boundaries. The cells evidence high velocities of cooling $\sim 10^6\text{--}10^7 \text{ K/s}$. The reason for the columnar crystallization is suggested to be the thermocapillary instability developing because of the temperature gradient in a liquid layer and forcing second-phase particles towards cell boundaries [43–45].

3.3. Mechanical Characteristics of Al–10.65Si–2.11Cu and Al–5.39Si–1.33Cu Irradiated with a High Current Pulsed Electron Beam

Such serious transformations in the structure of materials irradiated with the pulsed electron beam could not be irrelevant for mechanical properties. To characterize surface properties, the microhardness and wear parameter (a reciprocal of wear resistance) were in focus of studies and determined directly on the irradiated surface in all modification

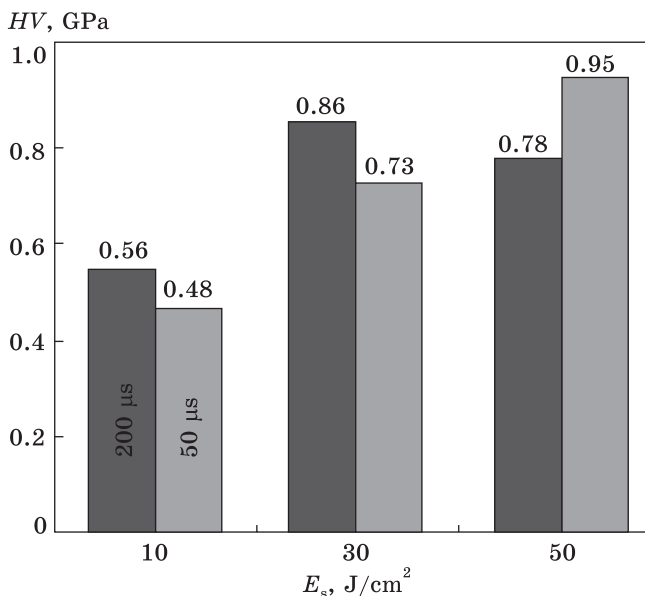


Fig. 24. The microhardness (HV) data vs. the energy density of electron beam (10, 30, 50 J/cm²) and pulse time (50 and 200 μ s). The HV value of the as delivered Al-5.39Si-1.33Cu alloy is 0.52 GPa

modes. The numerical data on the microhardness behaviour in the Al-5.39Si-1.33Cu alloy exposed to electron beam treatment are summarized in Fig. 24.

As seen in these data, the microhardness of the surface rises up to 0.95 GPa at the beam energy density of 50 J/cm² provided that a pulse time is set 50 μ s and a beam energy density kept increasing; this value is 83% higher than the same characteristic of the untreated material. Once a pulse time is 200 μ s, the surface microhardness is maximal (0.86 GPa) at the beam energy density of 30 J/cm², that is 65% higher than in the unprocessed material.

The outcomes of tribological testing are presented in Figure 25. It was observed the wear parameter decreases given the energy density of the electron beam is heightened in the irradiation with the electron beam of a pulse time of 200 μ s (Fig. 25, curve 1). The behaviour of the Al-5.39Si-1.33Cu alloy wear resistance is more complex if irradiated with a 50 μ s electron beam (Fig. 25, curve 2). In this case, the wear resistance shows a slightly decreasing trend at the beam energy density of 30 J/cm² and rises again if the energy density of the electron beam is set 50 J/cm². A maximal increase in wear resistance ($k = 0.37 \cdot 10^{-3}$ mm³/N·m) is registered for such treatment parameters as 50 J/cm², 200 μ s and amounts to 197%.

The data on the microhardness for the Al-10.65Si-2.11Cu alloy irradiated with the pulsed electron beam are summarized in Fig. 26. The bar diagram demonstrates a slight decrease of microhardness if the electron beam with the energy density of 10 J/cm² is applied; it may be

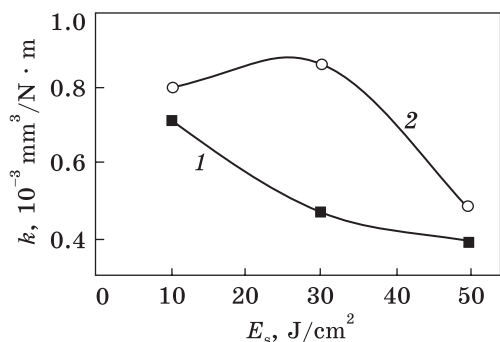


Fig. 25. The wear parameter k vs. the energy density of electron beam (10, 30, 50, J/cm²) and pulse time (50 and 200 μs). The parameter k of the as cast Al-5.39Si-1.33Cu alloy is $1.1 \cdot 10^{-3}$ mm³/N·m. 1 — pulse time if the electron beam 200 μs; 2 — 50 μs

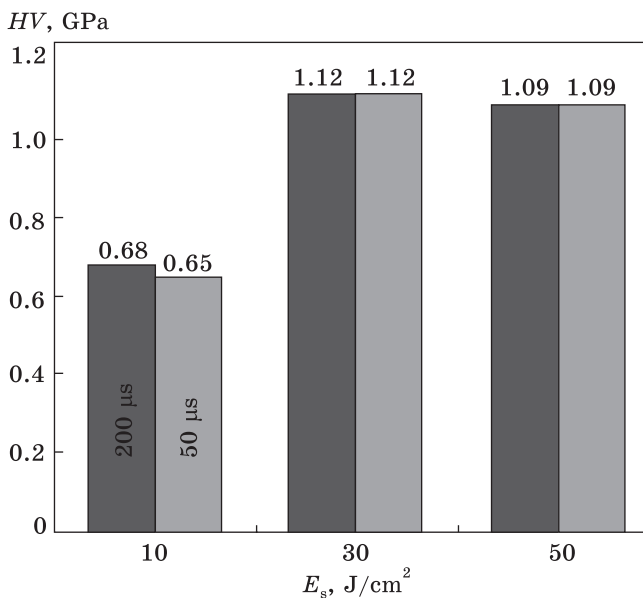


Fig. 26. The microhardness of Al-10.65Si-2.11Cu alloy surface irradiated with the pulsed electron beam (energy density and pulse time are different). The microhardness of the untreated as delivered material is as high as 0.72 GPa

related to the amount of energy transmitted from the beam to the surface, which is insufficient to melt the material and dissolve the intermetallic phase. Microhardness grows to 1.1 GPa on average if the energy density is heightening to 30–50 J/cm². Interestingly, there is no found relation between microhardness and pulse time of the electron beam. These transformations appear to be connected with the developing nanocrystalline structure of cell-type crystallization and the dissolving intermetallic phase, as shown by the SEM and TEM methods.

Studies revealed that the friction coefficient and wear intensity (a reciprocal of wear resistance) show a decreasing trend simultaneously to the microhardness growth in irradiated samples. If compared with the untreated material (parameters of tribological testing $P = 1$ N, $V = 25$ mm/s), the friction coefficient is ≈ 1.3 times lower and the wear

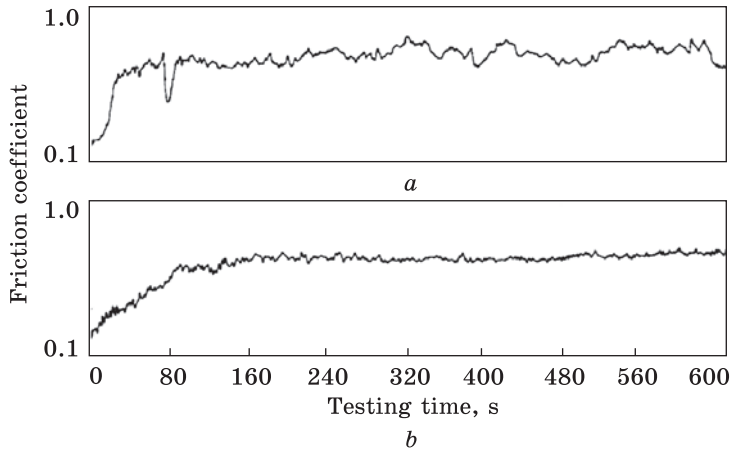


Fig. 27. Friction coefficient μ vs. time of tribological studies on samples of the as cast Al-10.65Si-2.11Cu alloy (a) and after irradiating with the intensive pulsed electron beam (b); parameters of tribological testing: $P = 1$ N, $V = 25$ mm/s [47]

intensity is ≈ 6.6 times lower. Figure 27 presents the diagrams characterizing the friction coefficient in tribological tests carried out under lower loads and higher velocity of tribological loading.

The tribological test data of the as cast (Fig. 27, curve a) and irradiated (Fig. 27, curve b) Al-10.65Si-2.11Cu alloy show the friction coefficient of the treated material begins to change in a stationary regime later (minimum 100 s), and the peak-to-peak amplitude of friction coefficient in the untreated sample is significantly higher ($\Delta\mu \geq 0.3$). It may be related to the solidification, breaking-off of hardening particles and to transfer of wear products from the zone of tribological contact onto the counterbody.

The correlation between results of tribological testing and microhardness is suggested, *i.e.*, wear resistance and microhardness increase if the energy density of the electron beam is heightened. Both facts evidence the surface hardening of the material.

4. Conclusions

This work was addressed to the element and phase composition, structure and mechanical properties of Al-10.65Si-2.11Cu and Al-5.39Si-1.33Cu alloy surfaces irradiated with the high intensity electron beam. Several conclusions were formulated in view of the research outcomes.

(i) The principal phases in the untreated alloys of interest include an aluminium-based solid solution, silicon and intermetallic compounds, *e.g.* a $\text{Fe}_2\text{Al}_9\text{Si}_2$ phase. In the Al-10.65Si-2.11Cu alloy, there is also a Cu_9Al_4 phase. As established, the treatment transforms the lattice pa-

parameter of Al–10.65Si–2.11Cu (an aluminium-based solid solution) and Al–5.39Si–1.33Cu (Al₁ and Al₂ phases) and changes the content of alloying elements in alloy surfaces.

(ii) The coherent scattering region is minimal (12.08 nm) in the Al–5.39Si–1.33Cu alloy (Si phase) for the beam energy density of 30 J/cm². Microdistortions of the Si crystal lattice demonstrate the opposite behaviour, increasing up to $3.576 \cdot 10^{-3}$ at 30 J/cm².

(iii) The coherent scattering region in the Al–10.65Si–2.11Cu alloy was determined for two AlSi and Si phases. The coherent scattering region was found to be maximal at 30 J/cm², irrespectively to the phase in focus, it is 92.92 nm for Si and 183.42 nm for AlSi. Microdistortions of the crystal lattice show an adverse trend, being a curve with the minimum observed at the beam energy density of 30 J/cm².

(iv) The modified layer has a nanocrystalline structure of cell-type crystallization. The thickness of this layer correlated with process parameters. Studies found rapid solidification cells are formed by an aluminium-based solid solution and range from 100 to 450 nm for Al–10.65Si–2.11Cu and from 40 to 100 nm for Al–5.39Si–1.33Cu. On the cell boundaries, there are silicon nanocrystallites; their sizes (5–10 nm) are identical for both alloys of interest.

(v) The microhardness of the Al–10.65Si–2.11Cu alloy displays the non-monotonous behaviour; it decreases at the beam energy density of 10 J/cm² and rises by 52.8%, if the energy density is set to 30–50 J/cm². The friction coefficient is ≈ 1.3 lower; the wear intensity is ≈ 6.6 times lower, provided that the microhardness is higher.

(vi) The nanocrystalline structure of cell-type crystallization, which is reinforced with silicon nanocrystals, makes harder the Al–5.39Si–1.33Cu alloy surface. A maximal increase of microhardness (0.95 GPa) is observed at the beam energy density of 50 J/cm² and a pulse time of 50 μ s. The wear resistance of the Al–5.39Si–1.33Cu alloy behaves non-monotonously: decreasing insignificantly at the beam energy density of 30 J/cm², but rising if the beam energy density is heightened. A maximal step up (197%) of the Al–5.39Si–1.33Cu wear resistance ($k = 0.37 \cdot 10^{-3}$ mm³/N·m) relates to the process parameters 50 J/cm² and 200 μ s.

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REFERENCES

1. V.Ya. Ivanchuk, *Tekhnol. Mashinostr.*, **1**: 73 (2001) (in Russian).
2. J.C. Benedyk, *Materials, Design and Manufacturing for Lightweight Vehicles* (Ed. P.K. Mallick) (Woodhead Publishing Series in Composites Science and Engineering: 2010), p. 79;
<https://doi.org/10.1533/9781845697822.1.79>

3. D. Brough and H. Jouhara, *Int. J. Thermofluids.*, 1–2: 100007 (2020);
<https://doi.org/10.1016/j.ijft.2019.100007>
4. Q.G. Wang, *Metall. Mater. Trans. A*, **34**: 2887 (2003);
<https://doi.org/10.1007/s11661-003-0189-7>
5. T. Wei, F. Yan, and J. Tian, *J. Alloys & Compd.*, **389**: 169 (2005);
<https://doi.org/10.1016/j.jallcom.2004.05.084>
6. M. Liu, J. Chen, Y. Lin, Z. Xue, H.J. Roven, and P.C. Skaret, *Prog. Nat. Sci. Mater. Int.*, **30**: 485 (2020);
<https://doi.org/10.1016/j.pnsc.2020.07.005>
7. C. Rochet, E. Andrieu, B. Arfaei, J.-P. Harouard, A. Laurino, T.C. Lowe, G. Odemer, and C. Blanc, *Int. J. Fatigue*, **140**: 105812 (2020);
<https://doi.org/10.1016/j.ijfatigue.2020.105812>
8. V. Andreyachshenko, *Mater. Lett.*, **254**: 433 (2019);
<https://doi.org/10.1016/j.matlet.2019.07.127>
9. K. Natori, H. Utsunomiya, and T. Tanaka, *J. Mater. Process. Technol.*, **240**: 240 (2017);
<https://doi.org/10.1016/j.jmatprotec.2016.09.022>
10. Z. Zribi, H.H. Ktari, F. Herbst, V. Optasanu, and N. Njah, *Mater Charact.*, **153**: 190 (2019);
<https://doi.org/10.1016/j.matchar.2019.04.044>
11. M. Ebrahimi, M.H. Shaeri, R. Naseri, and C. Gode, *Mater. Sci. Eng. A*, **731**: 569 (2018);
<https://doi.org/10.1016/j.msea.2018.06.080>
12. M. Hockauf, L.W. Meyer, B. Zillmann, M. Hietschold, S. Schulze, and L. Krüger, *Mater. Sci. Eng. A*, **503**: 167 (2009);
<https://doi.org/10.1016/j.msea.2008.02.051>
13. W. Xue, X. Shi, M. Hua, and Y. Li, *Appl. Surf. Sci.*, **253**: 6118 (2007);
<https://doi.org/10.1016/j.apsusc.2007.01.018>
14. N.Y. Dudareva, L.I. Zaynullina, and E.I. Ustimova, *Mater. Today Proc.*, **11**: 181 (2019);
<https://doi.org/10.1016/j.matpr.2018.12.128>
15. M. Mohedano, E. Matykina, R. Arrabal, B. Mingo, and A. Pardo, *Appl. Surf. Sci.*, **346**: 57 (2015);
<https://doi.org/10.1016/j.apsusc.2015.03.206>
16. K. Li, W. Li, G. Zhang, W. Zhu, F. Zheng, D. Zhang, and M. Wang, *J. Alloys & Compd.*, **790**: 650 (2019);
<https://doi.org/10.1016/j.jallcom.2019.03.217>
17. J. He, Q.Z. Cai, H.H. Luo, L. Yu, and B.K. Wei, *J. Alloys & Compd.*, **471**: 395 (2009);
<https://doi.org/10.1016/j.jallcom.2008.03.114>
18. P. He, L. Tang, G. Ma, H. Wang, S. Chen, M. Liu, S. Ding, Y. Bai, J. Tang, and D. He, *Appl. Surf. Sci.*, **530**: 147246 (2020);
<https://doi.org/10.1016/j.apsusc.2020.147246>
19. H. Bian, K. Aoyagi, Y. Zhao, C. Maeda, T. Mouri, and A. Chiba, *Addit. Manuf.*, **32**: 100982 (2020);
<https://doi.org/10.1016/j.addma.2019.100982>
20. W. Shaogang, H. Yan, and Z. Li, *Met. Mater. Eng.*, **47**, No. 7: 1973 (2018).
[https://doi.org/10.1016/S1875-5372\(18\)30167-X](https://doi.org/10.1016/S1875-5372(18)30167-X)
21. Yu.F. Ivanov, V.E. Gromov, D.V. Zagulyaev, S.V. Kononov, Yu.A. Rubannikova, and A. P. Semin, *Prog. Phys. Met.*, **21**, No. 3: 345 (2020);
<https://doi.org/10.15407/ufm.21.03.345>

22. Yu.F. Ivanov, V.E. Gromov, D.V. Zagulyaev, S.V. Kononov, Yu.A. Rubannikova, and A.P. Semin, *Prog. Phys. Met.*, **20**, No. 4: 634 (2019);
<https://doi.org/10.15407/ufm.20.04.634>
23. Yu.F. Ivanov, D.V. Zagulyaev, S.A. Nevskii, V.E. Gromov, V.D. Sarychev, and A.P. Semin, *Prog. Phys. Met.*, **20**, No. 3: 447 (2019);
<https://doi.org/10.15407/ufm.20.03.447>
24. Yu.F. Ivanov, V.E. Gromov, S.V. Kononov, D.V. Zagulyaev, E.A. Petrikova, and A.P. Semin, *Prog. Phys. Met.*, **19**, No. 2: 195 (2018);
<https://doi.org/10.15407/ufm.19.02.195>
25. V. E. Kormyshev, V.E. Gromov, Yu.F. Ivanov, and S.V. Kononov, *Usp. Fiz. Met.*, **18**, No. 2: 111 (2017) (in Russian);
<https://doi.org/10.15407/ufm.18.02.111>
26. A.D. Teresov, V.V. Shugurov, Yu.F. Ivanov, Yu.A. Denisova, E.A. Petrikova, and N.N. Koval', *Izvestiya Vuzov. Fizika*, **58**: 112 (2015) (in Russian).
27. V.A. Shulov, A.N. Gromov, D.A. Teryaev, and V.I. Engel'ko, *Izvestiya Vuzov. Poroshkovaya Metallurgiya i Funktsional'nyye Pokrytiya*, No. 1: 38 (2015) (in Russian);
<https://doi.org/10.17073/1997-308X-2015-1-38-48>
28. V.A. Shulov, V.I. Engelko, A.N. Gromov, D.A. Teryaev, O.A. Bytsenko, and G.G. Shivaryants, *Physics and Chemistry of Materials Treatment*, No. 5: 22 (2015) (in Russian).
29. T.C. Zhang, K.M. Zhang, J.X. Zou, P. Yan, H.Y. Yang, L.X. Song, and X. Zhang, *J Alloys & Compd.*, **788**: 231 (2019);
<https://doi.org/10.1016/j.jallcom.2019.02.130>
30. X. Zhang, K. Zhang, J. Zou, P. Yan, L. Song, and Y. Liu, *Vacuum*, **173**: 109132 (2020);
<https://doi.org/10.1016/j.vacuum.2019.109132>
31. L. Wang, K. Wang, N. Erkan, Y. Yuan, J. Chen, B. Nie, F. Li, and K. Okamoto, *Appl. Surf. Sci.*, **511**: 145555 (2020);
<https://doi.org/10.1016/j.apsusc.2020.145555>
32. P. Yu, M. Yan, D. Tomus, C.A. Brice, C.J. Bettles, B. Muddle, and M. Qian, *Mater. Charact.*, **143**: 43 (2018);
<https://doi.org/10.1016/j.matchar.2017.09.005>
33. B. Gao, N. Xu, and P. Xing, *Mater Lett.*, **237**: 180 (2019);
<https://doi.org/10.1016/j.matlet.2018.11.054>
34. H. Li, W. Xu, Z. Wang, B. Fang, R. Song, and Z. Zheng, *Mater. Sci. Eng. A*, **650**: 254 (2016);
<https://doi.org/10.1016/j.msea.2015.10.051>
35. P. Yan, T. Grosdidier, X. Zhang, and J. Zou, *Mater. Design*, **159**: 1 (2018);
<https://doi.org/10.1016/j.matdes.2018.08.033>
36. P. Yan, J. Zou, C. Zhang, and T. Grosdidier, *Appl. Surf. Sci.*, **504**: 144382 (2020);
<https://doi.org/10.1016/j.apsusc.2019.144382>
37. B. Gao, Y. Hao, W.F. Zhuang, G.F. Tu, W.X. Shi, S.W. Li, S.Z. Hao, C. Dong, and M.C. Li, *Phys. Procedia*, **18**: 187 (2011);
<https://doi.org/10.1016/j.phpro.2011.06.079>
38. J. An, X.X. Shen, Y. Lu, Y.B. Liu, R.G. Li, C.M. Chen, and M.J. Zhang, *Surf. Coatings Technol.*, **200**, Nos. 18–19: 5590 (2006);
<https://doi.org/10.1016/j.surfcoat.2005.07.106>
39. N.N. Koval' and Y.F. Ivanov, *Russ Phys. J.*, **51**: 505 (2008);
<https://doi.org/10.1007/s11182-008-9073-7>

40. D. Zaguliaev, S. Konovalov, Y. Ivanov, and V. Gromov, *Appl. Surf. Sci.*, **498**: 143767 (2019);
<https://doi.org/10.1016/j.apsusc.2019.143767>
41. B. Daemi, R. Tomkowski, and A. Archenti, *CIRP Ann.*, **69**, No. 1: 433 (2020);
<https://doi.org/10.1016/j.cirp.2020.03.022>
42. A.P. Babichev, O.N. Babushkina, and A.M. Bratkovsky, *Fizicheskie Velichiny: Spravochnik* [Physical Values: Reference Book] (Moscow: Energoatomizdat: 1991) (in Russian).
43. D. Zaguliaev, I. Chumachkov, Y. Ivanov, A. Abaturova, A. Ustinov, and A. Semin, *2020 7th International Congress on Energy Fluxes and Radiation Effects (EFRE) (14–26 Sept. 2020, Tomsk, Russia)* (IEEE: 2020), p. 737;
<https://doi.org/10.1109/EFRE47760.2020.9242140>
44. S. Konovalov, D. Zaguliaev, Y. Ivanov, V. Gromov, and A. Abaturova, *J. Mater. Res. Technol.*, **9**, No. 3: 5591 (2020);
<https://doi.org/10.1016/j.jmrt.2020.03.083>
45. D. Zaguliaev, S. Konovalov, Y. Ivanov, A. Abaturova, and A. Leonov, *Arch. Foundry Eng.*, **20**, No. 3: 92 (2020);
<https://doi.org/10.24425/afe.2020.133336>
46. V.V. Shlyarov, D.V. Zaguliaev, A.A. Abaturova, A.A. Leonov, and A.M. Ustinov, *Basic Problems of Material Science*, **17**, No. 1: 32 (2020) (in Russian);
<https://doi.org/10.25712/ASTU.1811-1416.2020.01.005>
47. D.V. Zagulyaev, V.E. Gromov, S.V. Konovalov, A.M. Glezer, S.V. Panin, and Yu.F. Ivanov, *Russ. Metall.*, **2019**: 981 (2019);
<https://doi.org/10.1134/S0036029519100318>

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ВПЛИВ ВИСОКОІНТЕНСИВНОГО ЕЛЕКТРОННОГО ПУЧКА НА КРИСТАЛІЧНУ БУДОВУ, ФАЗОВИЙ СКЛАД І ВЛАСТИВОСТІ СТОПІВ Al–Si З РІЗНИМ ВМІСТОМ СИЛІЦІЮ

Роботу спрямовано на вивчення елементного та фазового складу, мікроструктурної еволюції, параметра та мікроспотворень кристалічної ґратниці та розміру ділянки когерентного розсіяння у стопах Al–10,65Si–2,11Cu і Al–5,39Si–1,33Cu, оброблюваних високоінтенсивним електронним пучком. Методами рентгенофазової аналізи встановлено, що у вихідному стані основними фазами досліджуванних стопів є твердий розчин на основі алюмінію, силіцій та інтерметаліди, а також наявна фаза складу Fe₂Al₉Si₂. У стопі Al–10,65Si–2,11Cu додатково виявлено фазу Cu₉Al₄. Опромінення стопів імпульсним електронним пучком супроводжується зміною параметра кристалічної ґратниці Al–10,65Si–2,11Cu (твердий розчин на основі алюмінію) та Al–5,39Si–1,33Cu (фази Al₁ і Al₂). Ймовірно, причиною зміни параметра кристалічної ґратниці в стопах Al–10,65Si–2,11Cu та

Al-5,39Si-1,33Cu є зміна концентрації легувальних елементів у твердому розчині даних фаз. Встановлено, що при густині пучка електронів у 30 і 50 Дж/см² у модифікованому шарі спостерігається розчинення силіцію та інтерметалідів. Сучасні методи фізичного матеріалознавства уможливили встановити, що в результаті опромінення поверхні матеріалу формується шар з нанокристалічною структурою коміркової кристалізації. Товщина модифікованого шару варіюється залежно від параметрів електронно-пучкового оброблення й сягає максимального значення у 90 мкм за густини енергії у 50 Дж/см². За даними сканувальної (СЕМ) та просвітлювальної (ПЕМ) електронної мікроскопії частинки силіцію розташовуються на межах комірок. Подібні зміни структурно-фазових станів матеріалів позначаються й на механічних характеристиках. У якості характеристики поверхневих шарів використовували мікротвердість, параметр зношення та коефіцієнт тертя, значення яких визначали безпосередньо на поверхні опромінення, для всіх варіантів модифікування. Показано, що при оброблянні поверхні матеріалу інтенсивним імпульсним електронним пучком відбувається збільшення зносостійкості та мікротвердості стопів Al-10,65Si-2,11Cu і Al-5,39Si-1,33Cu.

Ключові слова: вилітні алюмінієві стопи, електронно-пучкове оброблення, мікроструктура, механічні властивості, структурно-фазові перетворення, сканувальна електронна мікроскопія, просвітлювальна електронна мікроскопія, мікротвердість, трибологічні випробування.